



Origin of the Elements: Evolution and Nucleosynthesis in Massive Stars

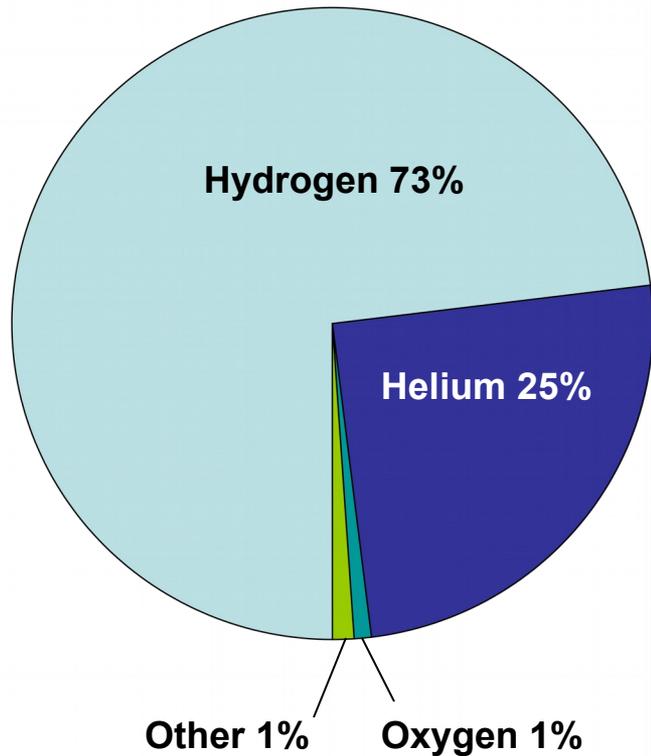
Alexander Heger

Overview

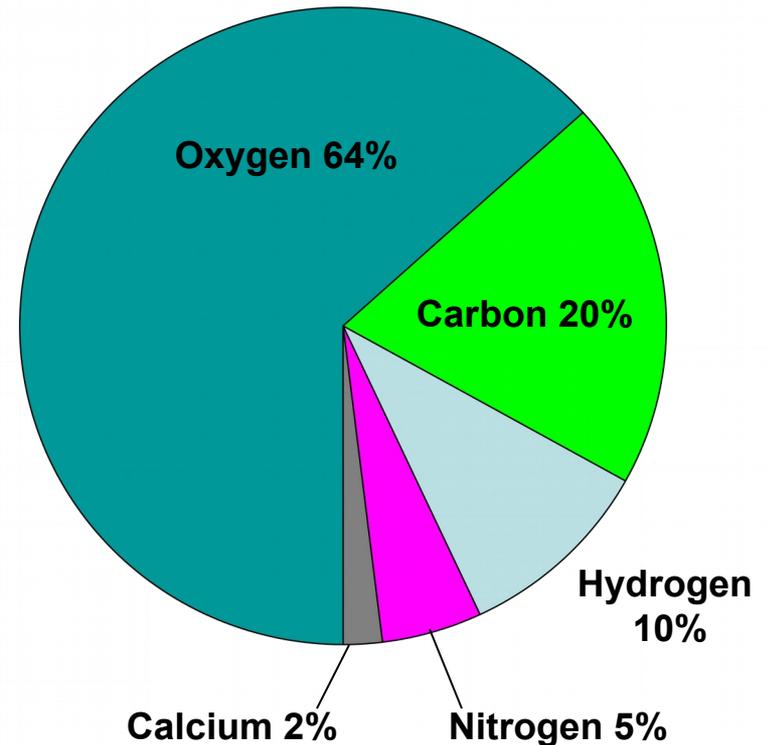
- **Presupernova Evolution**
- **Nucleosynthesis**
- **Varieties of Stellar Deaths**

Abundance by Weight

Universe

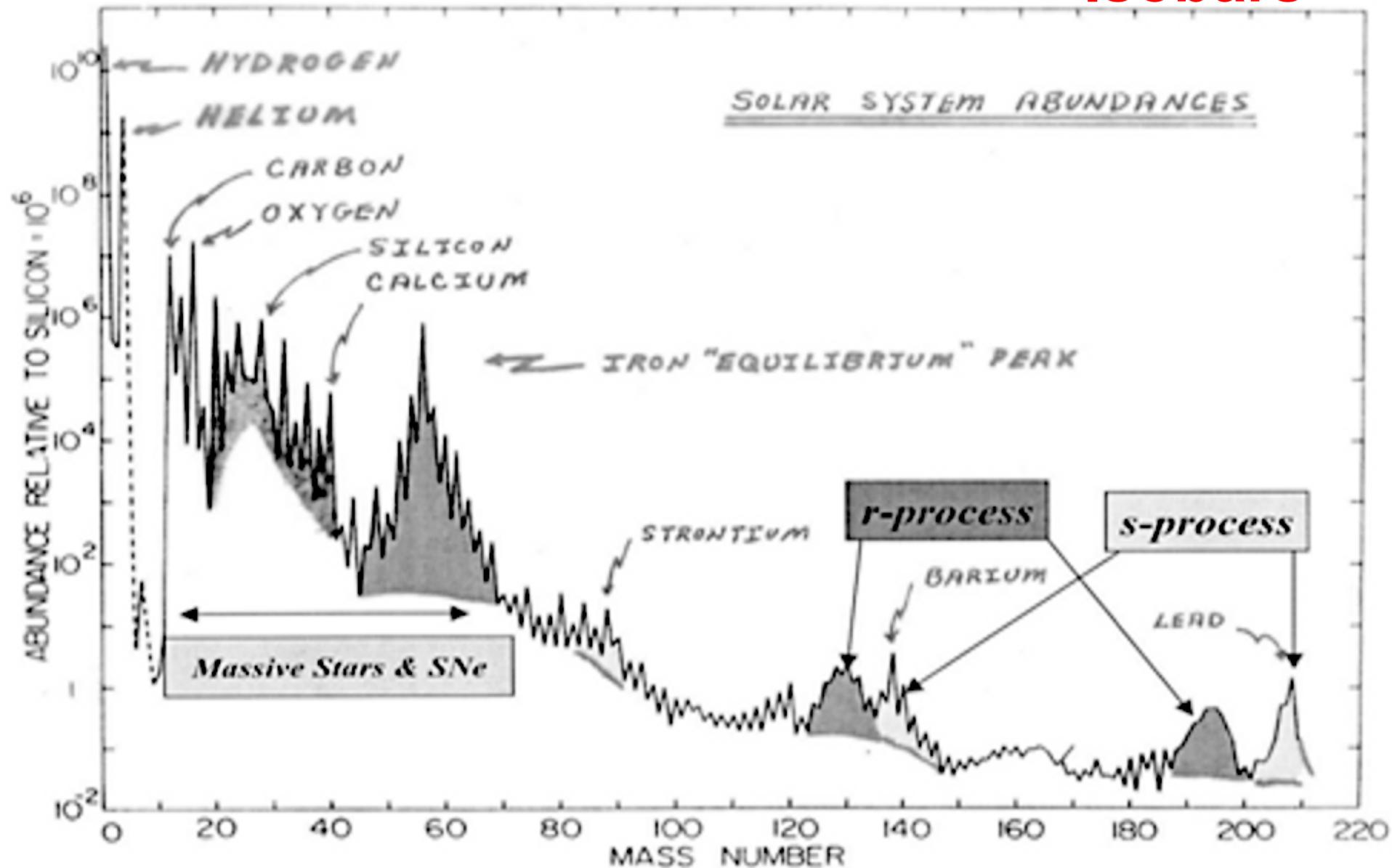


Humans



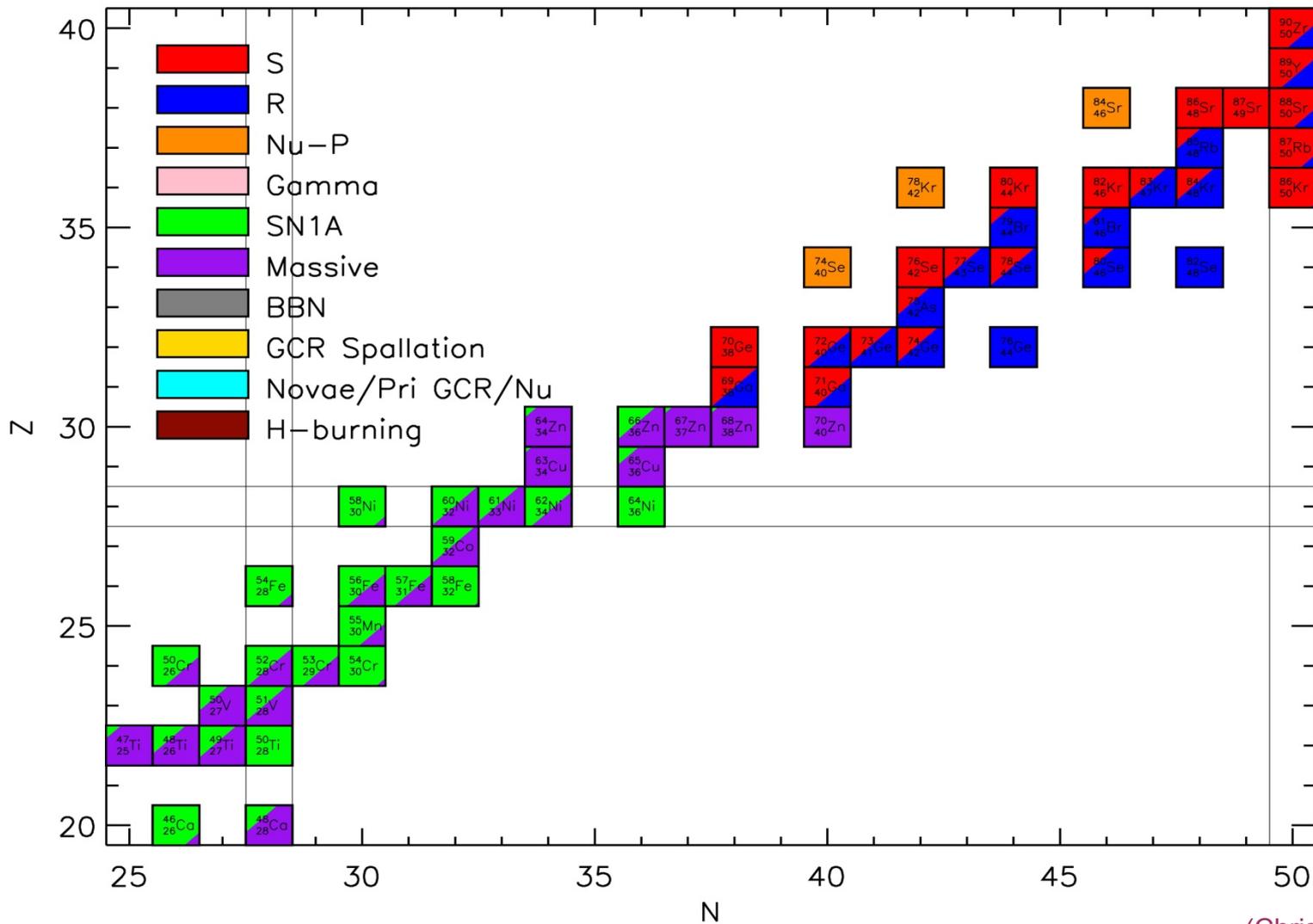
~~“Cosmic” Abundances of the Elements~~

Isobars



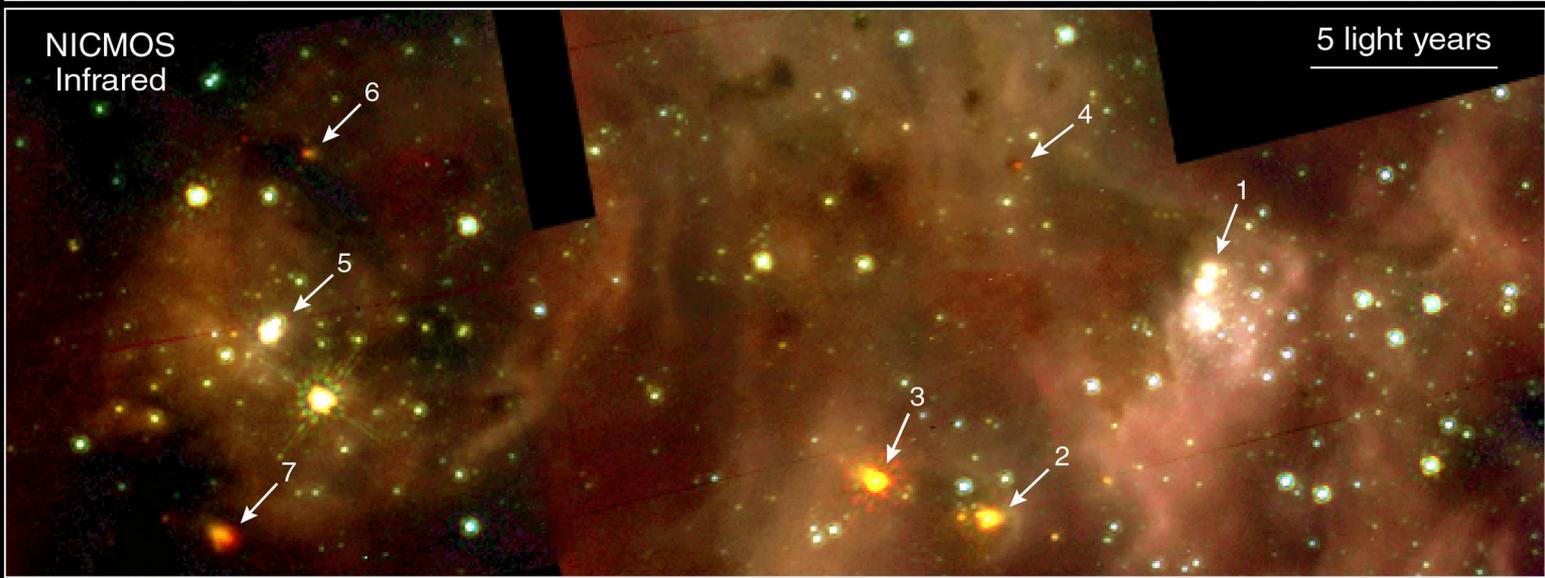
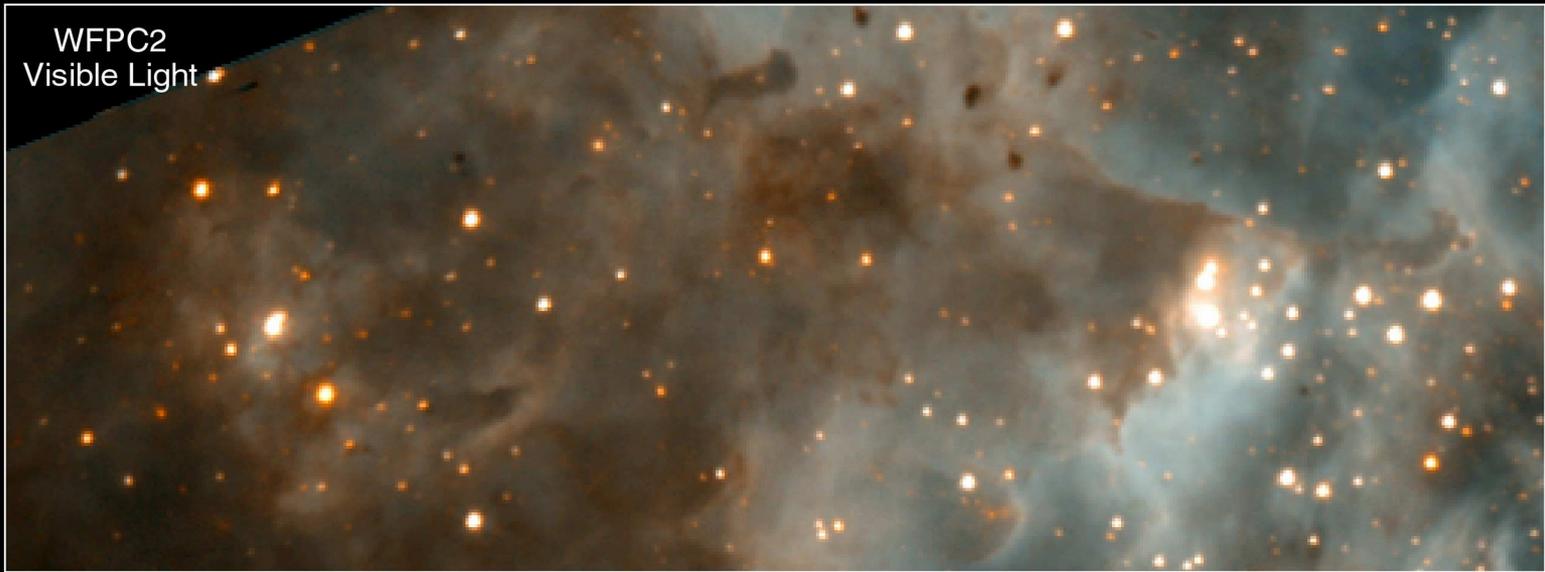
Isotope Decomposition by Process

(for illustrative purposes only)



Chapter One:

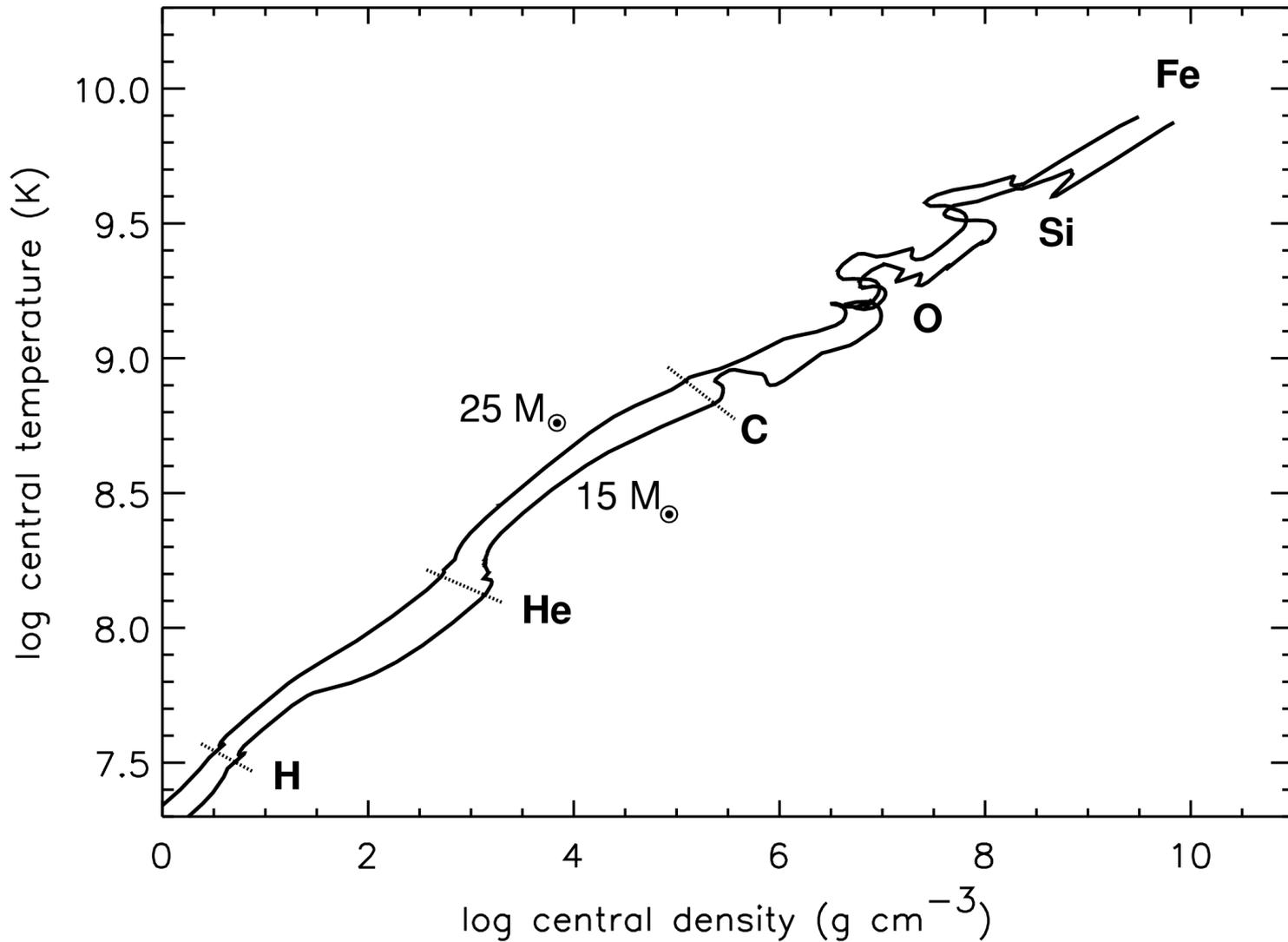
Pre-Supernova Evolution and Nucleosynthesis



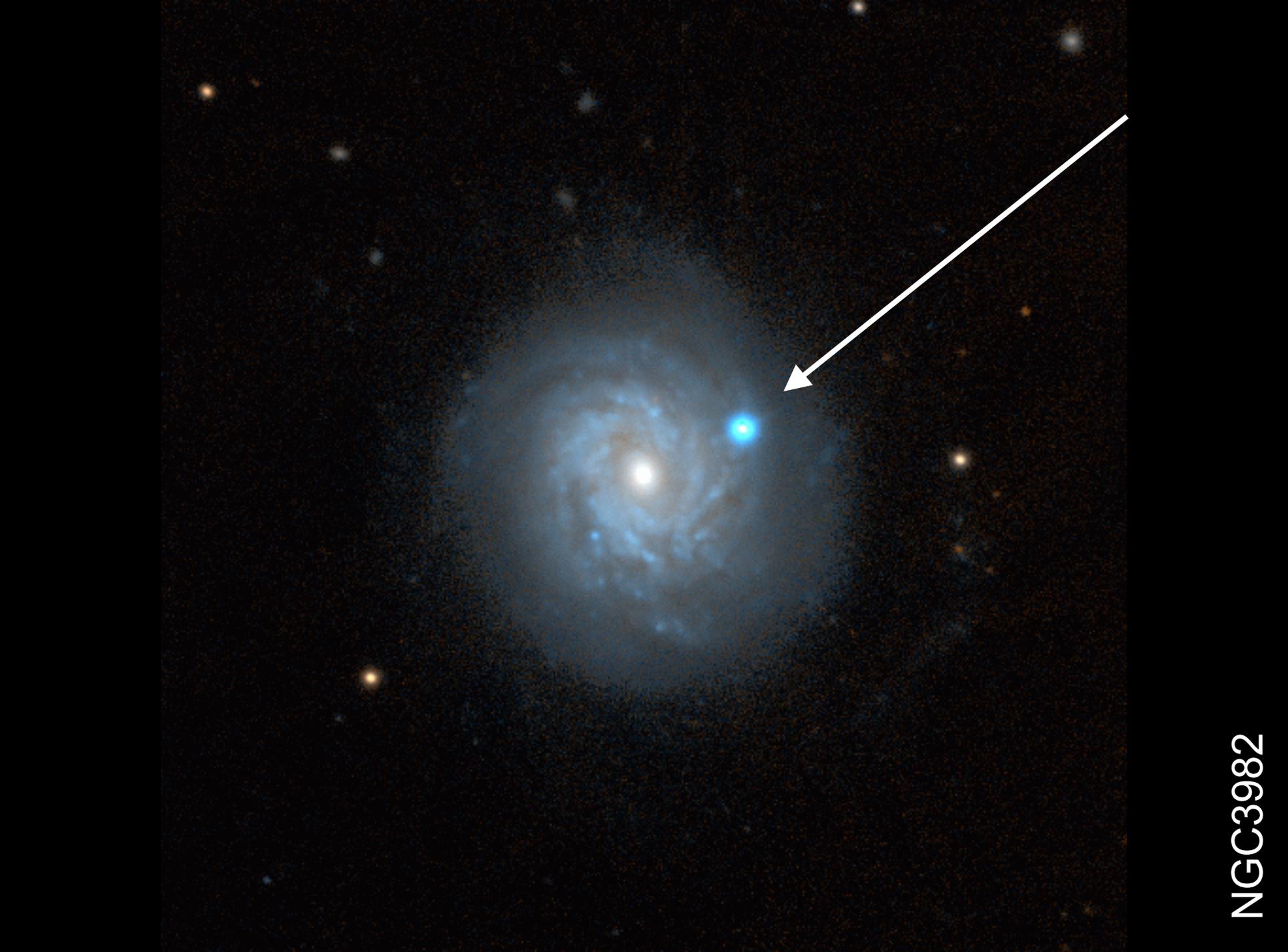
30 Doradus Details
Hubble Space Telescope • WFPC2 • NICMOS

PRC99-33b • STScI OPO • N. Walborn (STScI), R. Barbá (La Plata Observatory) and NASA

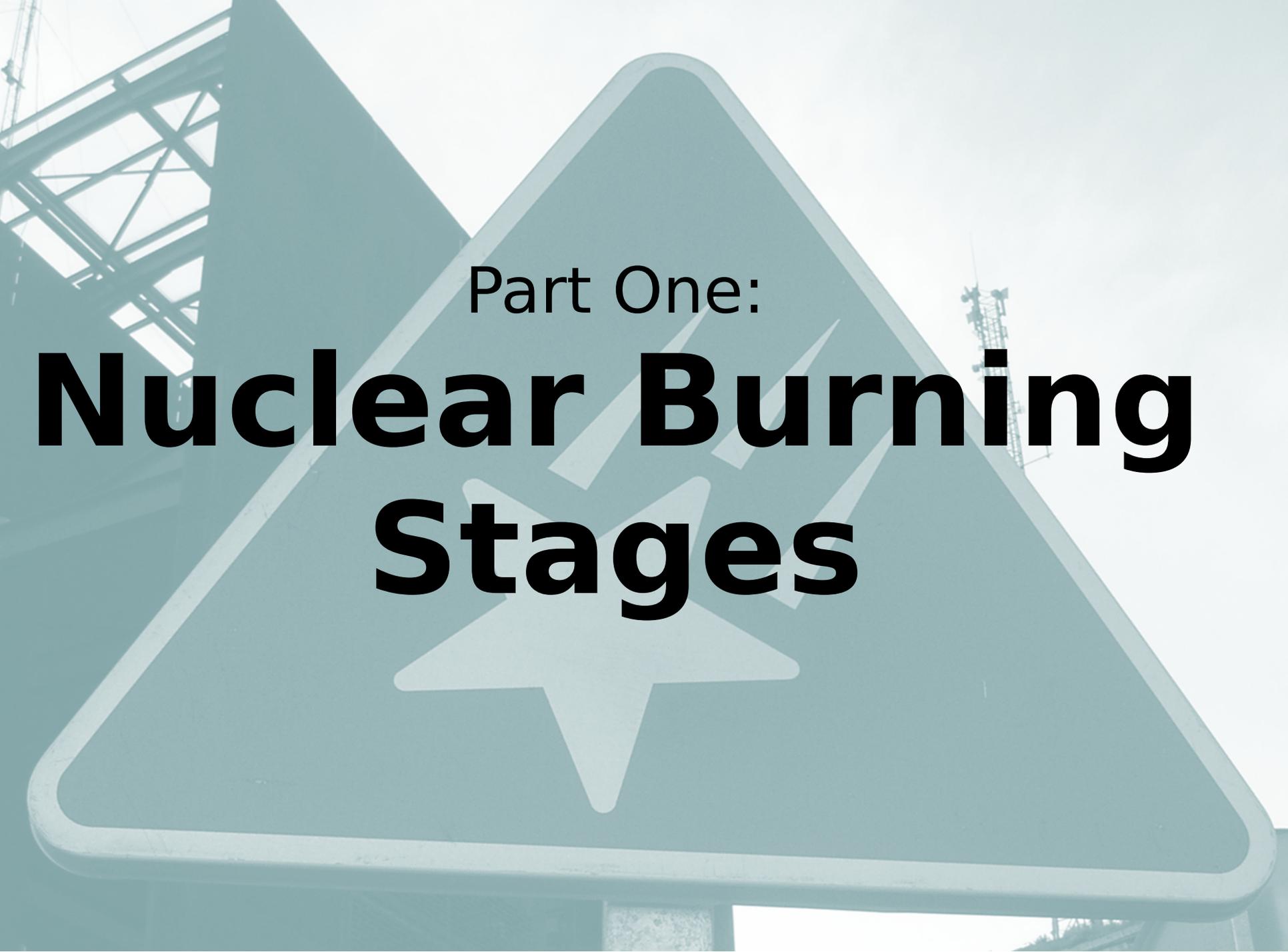
Once formed, the evolution of a star is governed by gravity:
continuing contraction
to higher central densities and temperatures



Evolution of
central
density and
temperature
of $15 M_{\odot}$
and $25 M_{\odot}$
stars



NGC3982

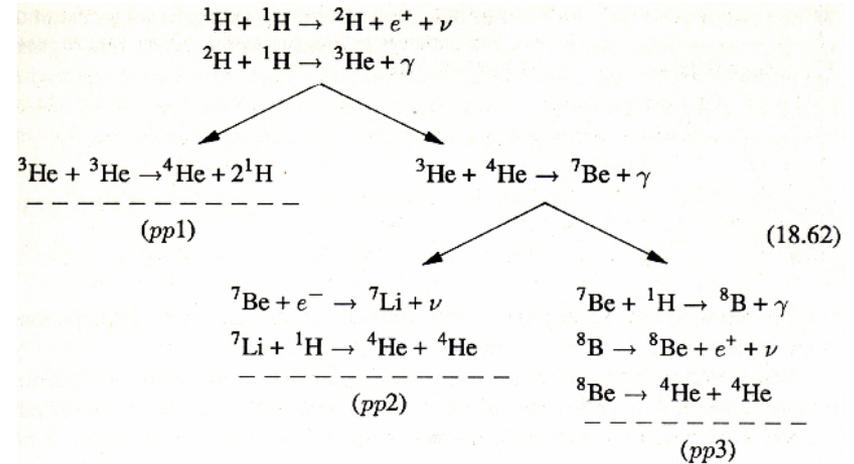
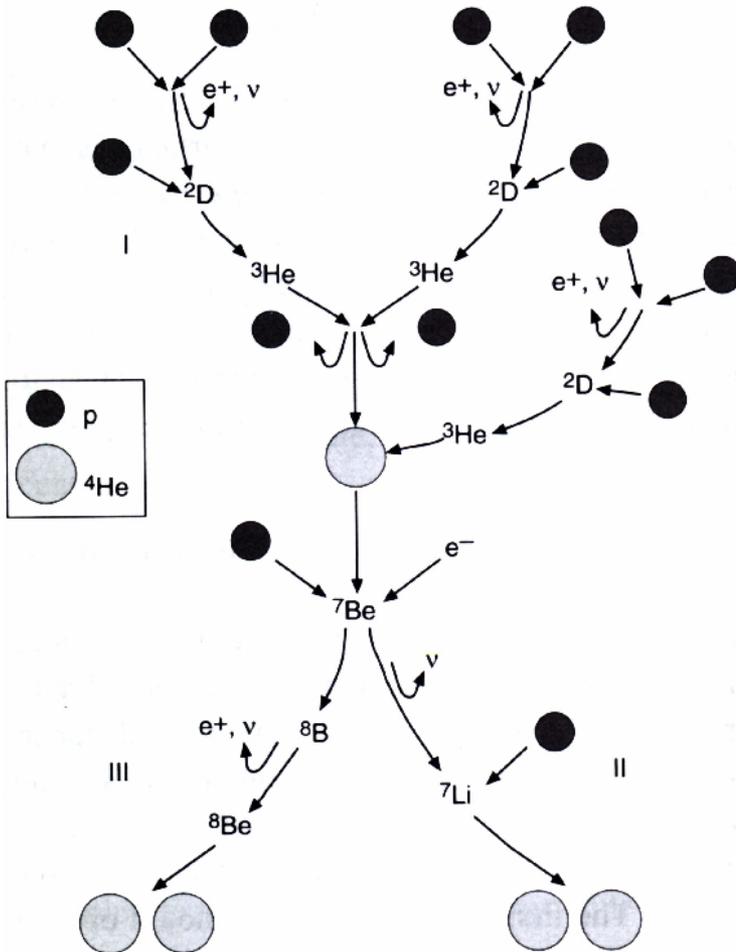


Part One:

Nuclear Burning Stages

Hydrogen-Burning: pp Chains

Hydrogen burning



Energy release:

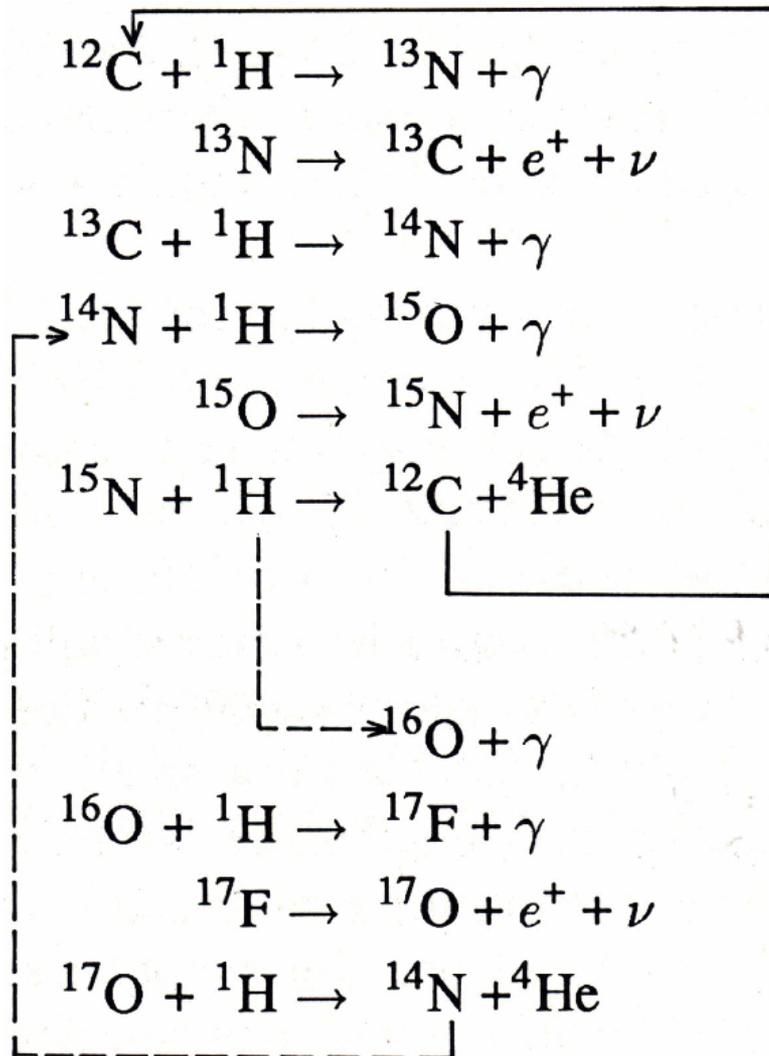
$$Q(pp1) = 26.20 \text{ MeV}$$

$$Q(pp2) = 25.67 \text{ MeV}$$

$$Q(pp3) = 19.20 \text{ MeV}$$

$$\text{Reaction rate: } \langle \sigma v \rangle \propto T^4$$

Hydrogen Burning: CNO Bi-Cycle



Energy release:

$$Q(\text{CNO}) = 24.97 \text{ MeV}$$

Reaction rate: $\langle \sigma v \rangle \propto T^{16}$

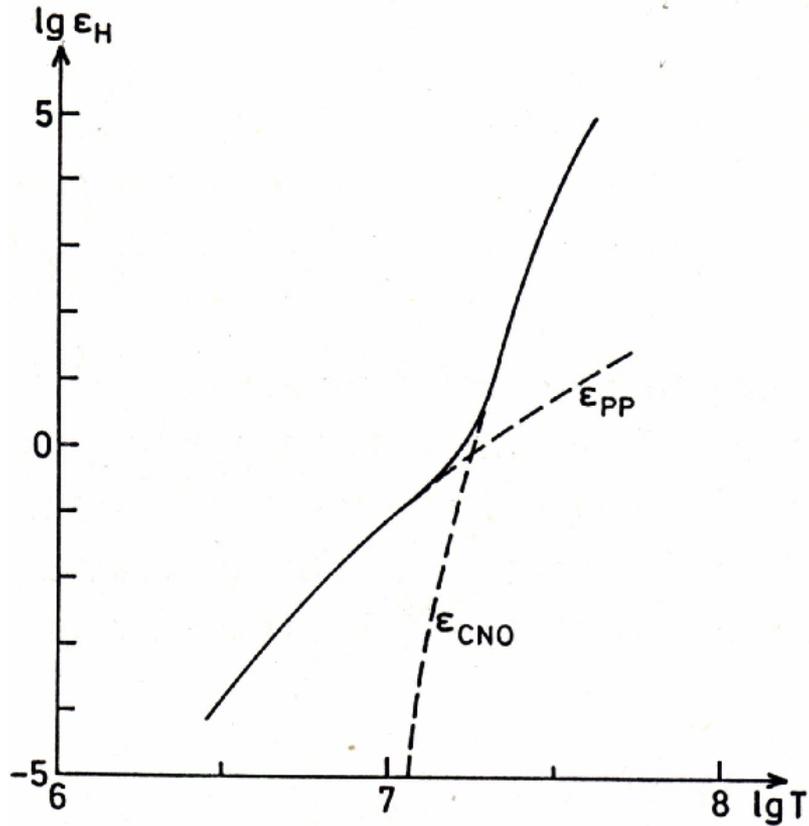
Branching:

CNO-1 : CNO-2 \sim 10,000 : 1

Hydrogen Burning: CNO Bi-Cycle

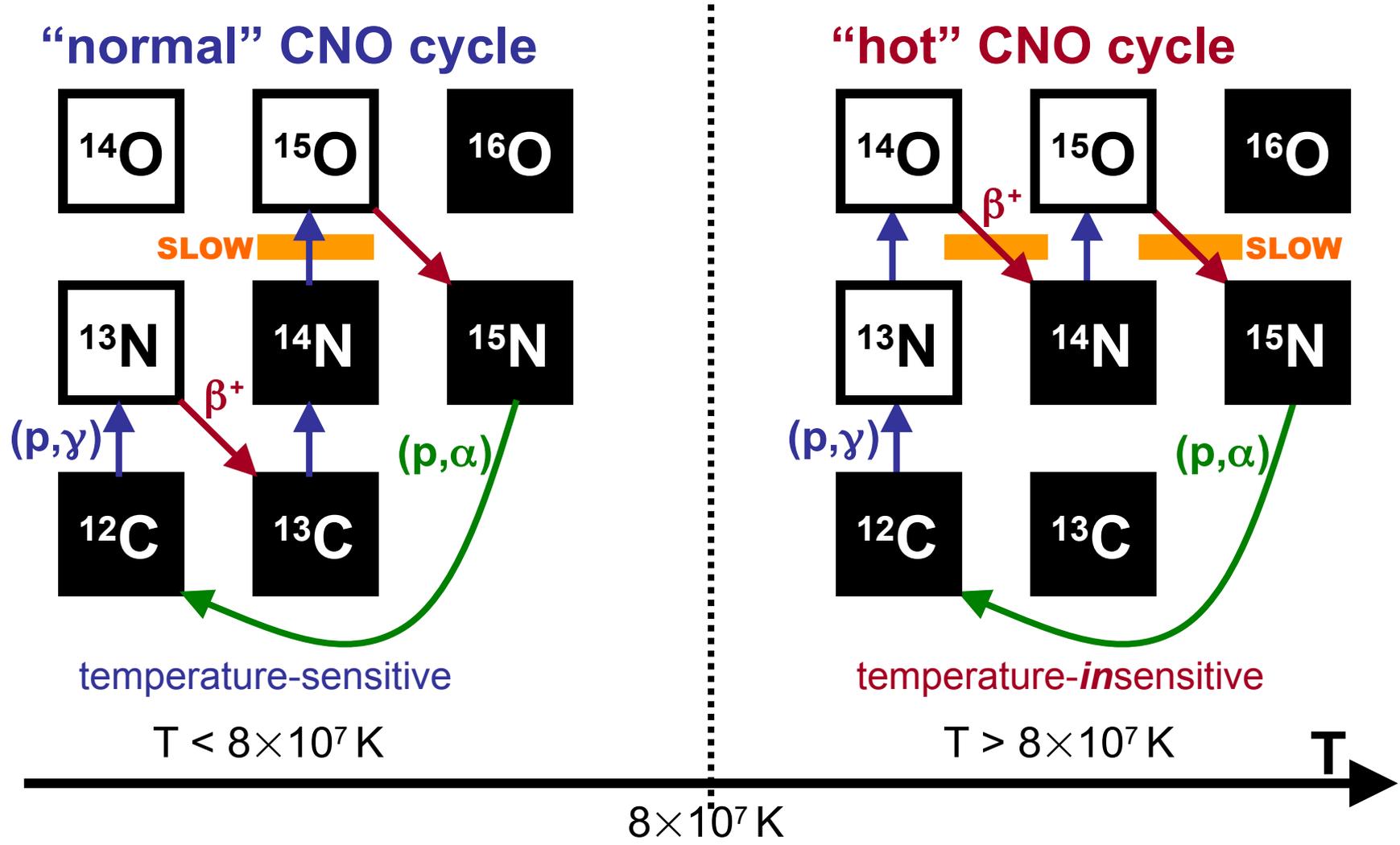
- Usually the beta-decays are fast compared to the capture reactions, (p, γ) .
- ^{14}O : $\tau_{1/2} = 70 \text{ sec}$
- ^{15}O : $\tau_{1/2} = 122 \text{ sec}$
- ^{13}N : $\tau_{1/2} = 10 \text{ min}$
- ^{17}F : $\tau_{1/2} = 64 \text{ sec}$
- ^{18}O : $\tau_{1/2} = 110 \text{ min}$
- $^{14}\text{N}(p, \gamma)^{15}\text{O}$ usually is the slowest “bottleneck” reaction.
- CNO cycle burning converts most CNO isotopes into ^{14}N .

Competition of Hydrogen-Burning Modes



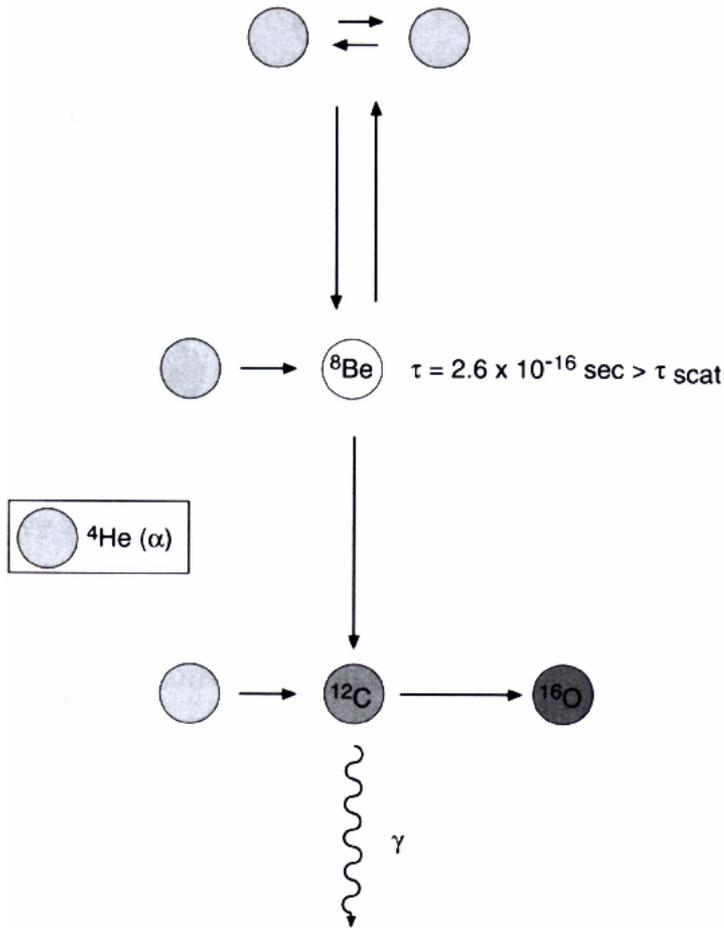
Transition from pp-chains
in low-mass stars (low T)
to CNO chains
in high-mass stars (high T)

Hydrogen Burning by CNO Cycle



time for an eddy to burn its hydrogen content by **hot** CNO cycle $\tau_H = 11 \text{ h} \left(\frac{0.02}{Z} \right) \left(\frac{X_0}{0.7} \right)$

Helium Burning

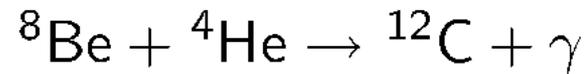


Step 1:



Built up equilibrium abundance of ^8Be
 Lifetime of ^8Be is only $2.6 \times 10^{-16} \text{ s}$!

Step 2:



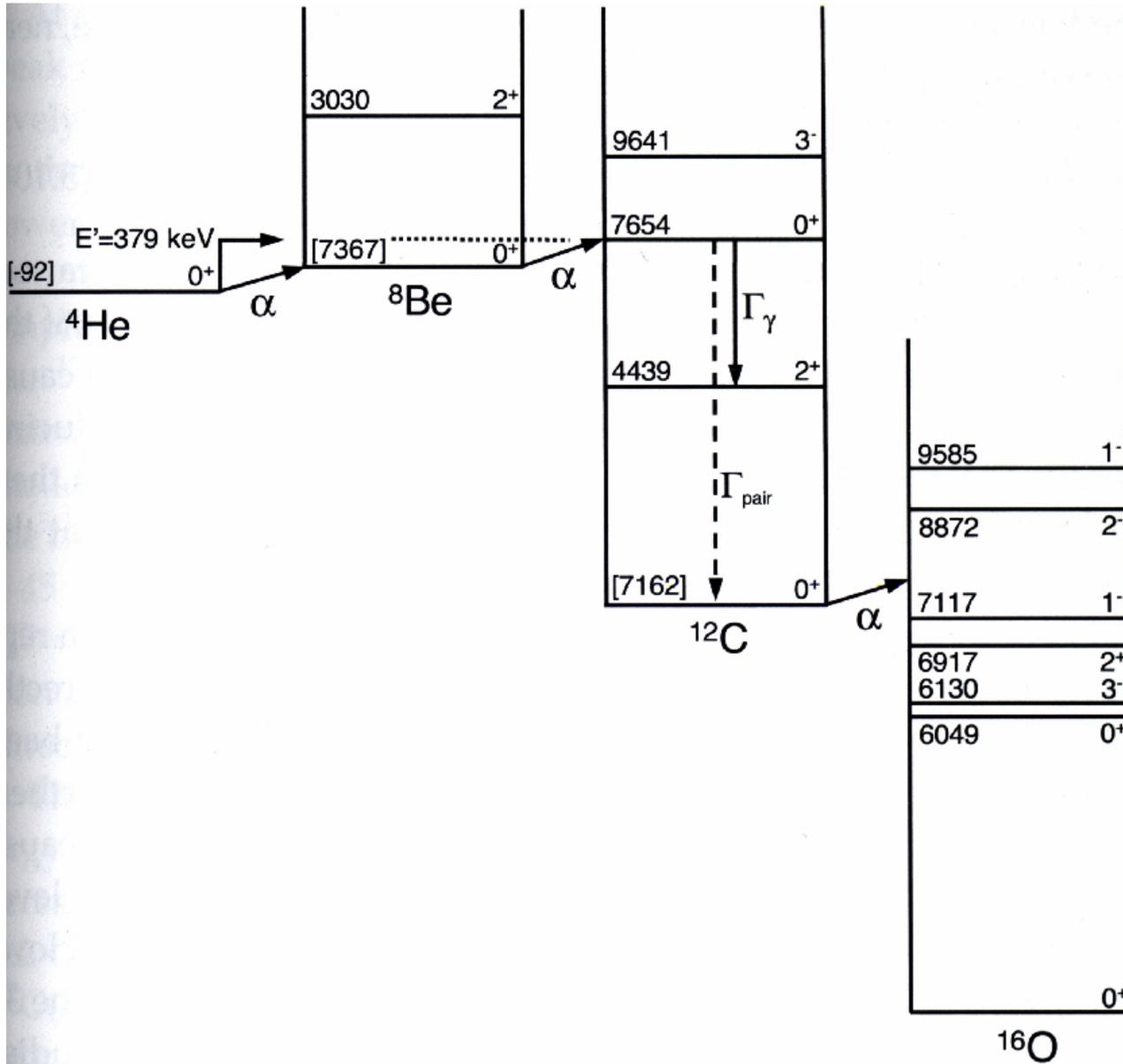
$$Q_{3\alpha} = 7.275 \text{ MeV}$$

$$\langle \sigma v \rangle \propto \rho^2 T^{40}$$

$$T = 10^8 \text{ K} \quad \Rightarrow \quad n(^8\text{Be}) : n(^4\text{He}) = 1 : 10^9$$

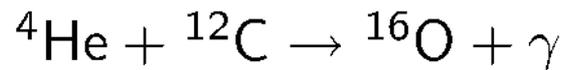
$$\rho = 10^5 \text{ g cm}^{-3}$$

Helium Burning Level Scheme



Additional Helium Burning Reactions

Oxygen Production



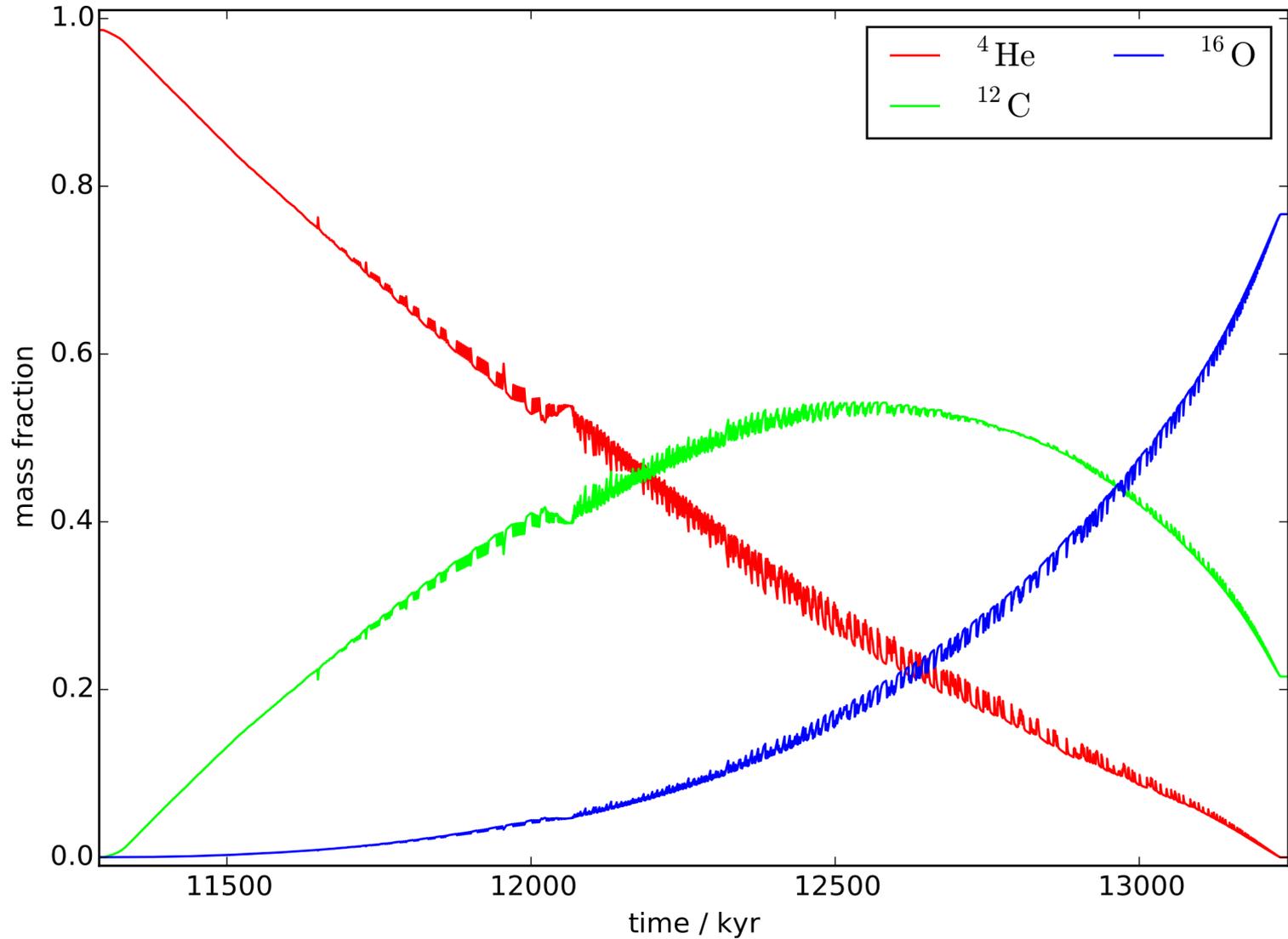
$$Q = 7.162 \text{ MeV}$$

$$\langle \sigma v \rangle \propto \rho T^{40}$$

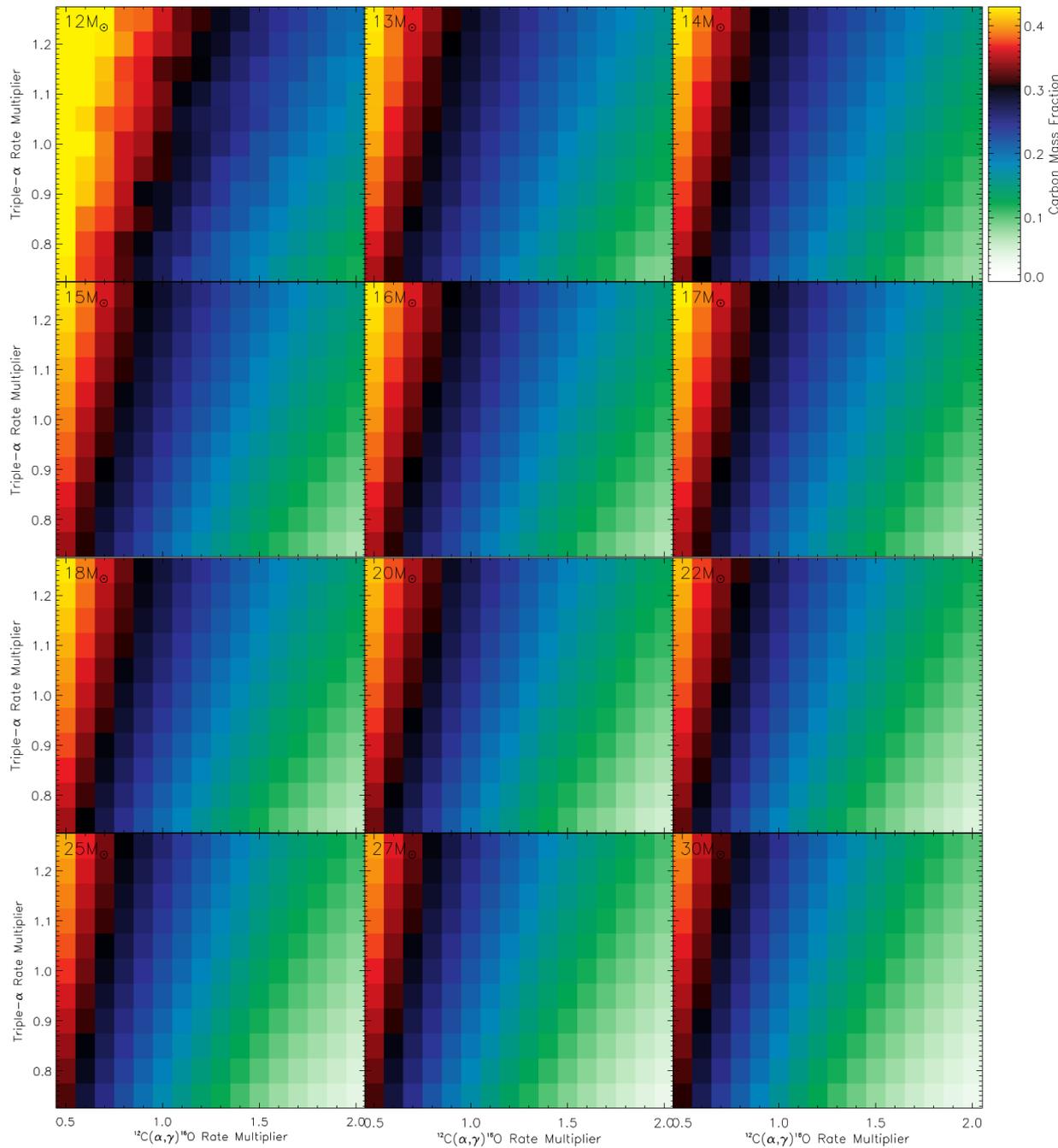
The final abundance of carbon is set by the competition of 3α and ${}^{12}\text{C}(\alpha, \gamma){}^{16}\text{O}$ reactions;

The production of ${}^{16}\text{O}$ can only start when a sufficient amount of ${}^{12}\text{C}$ has been made.

Competition of Helium Burning Reactions



^{12}C Production as a function of $^{12}\text{C}(\alpha,\gamma)$ and 3α reaction rates

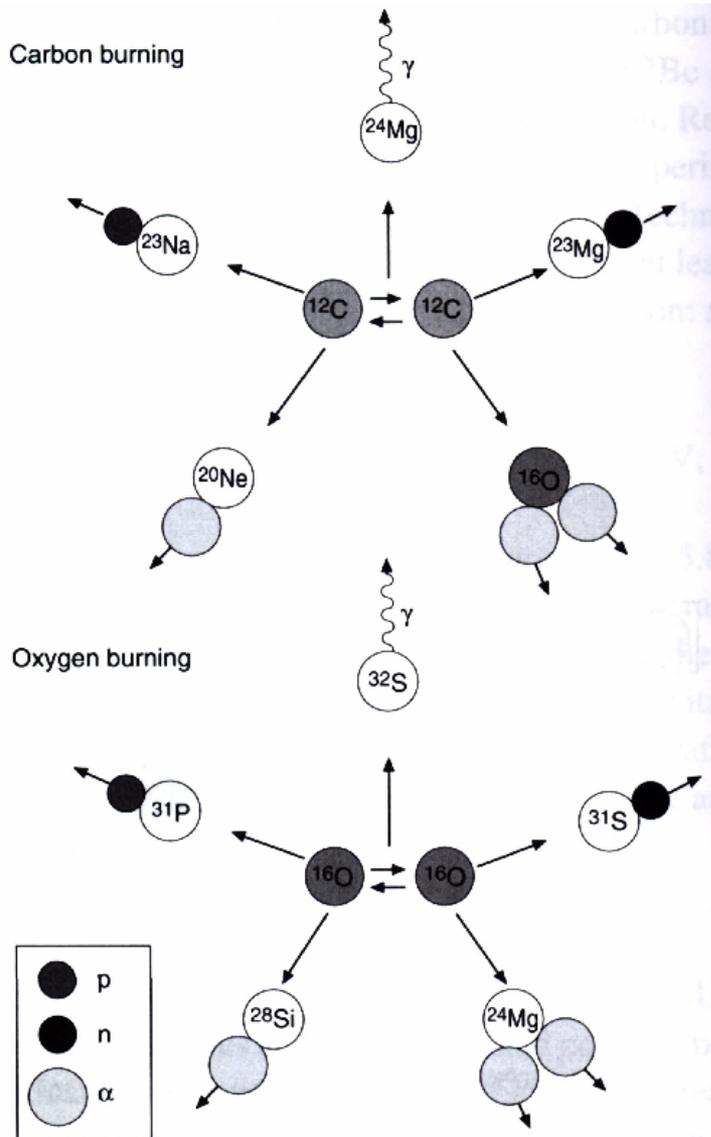


Carbon mass fraction at the end of helium burning depends the reaction rates and the mass of the star

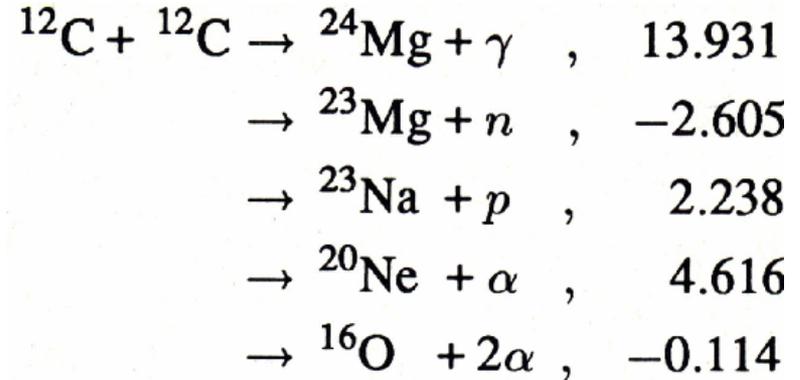
~2000 stellar models

(West+ 2013)

Carbon and Oxygen Burning

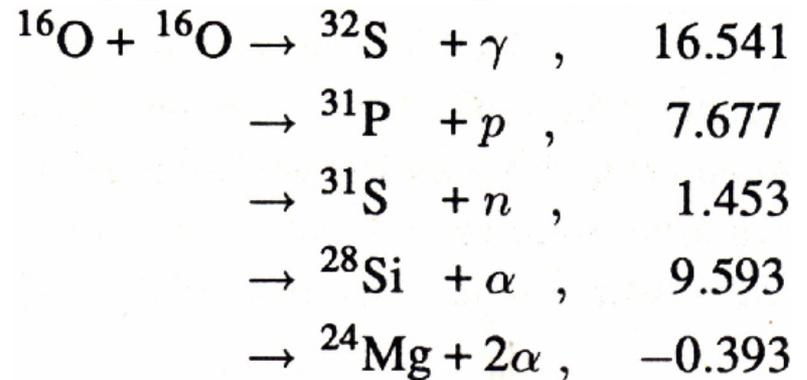


Carbon Burning



Average $Q = 13 \text{ MeV}$

Oxygen Burning



Average $Q = 16 \text{ MeV}$

Neutrino losses from electron/positron pair annihilation

- Important for carbon burning and beyond

- For $T > 10^9$ K (about 100 keV), occasionally:



and usually



but sometimes

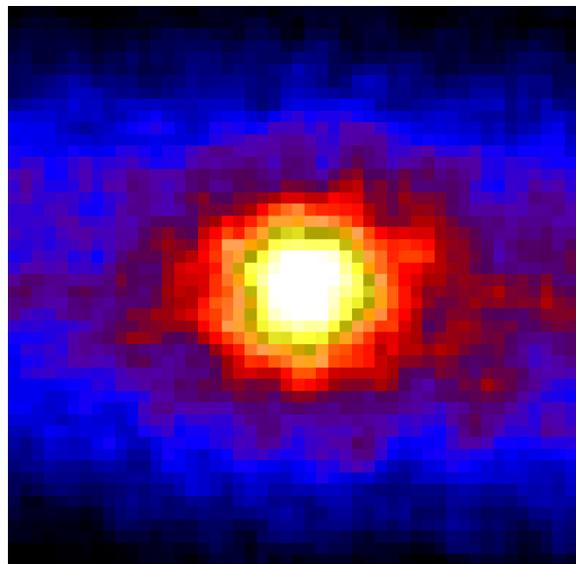


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- The neutrinos exit the stars at the speed of light while the e^+ , e^- , and the γ 's all stay trapped.

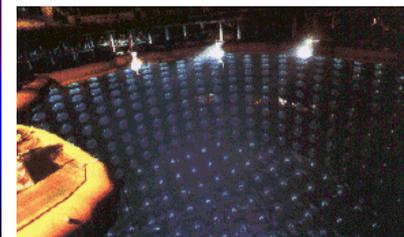
- This is an important energy loss with

$$\epsilon_\nu \approx -10^{15} (T/10^9\text{K})^9 \text{ erg g}^{-1} \text{ s}^{-1}$$

- For carbon burning and beyond, each burning stage gives about the same energy per nucleon, thus the lifetime goes down as T^{-9}



The sun as seen by Kamiokande



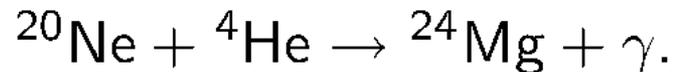
Neon Burning

Neon burning proceeds by a combination of photo-disintegrations and α captures:

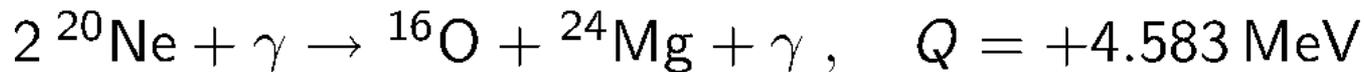


This reaction dominates over the inverse reaction known from helium burning for $T > 1.5 \times 10^9 \text{ K}$.

Subsequently, the ${}^4\text{He}$ is captured on another ${}^{20}\text{Ne}$ nucleus:



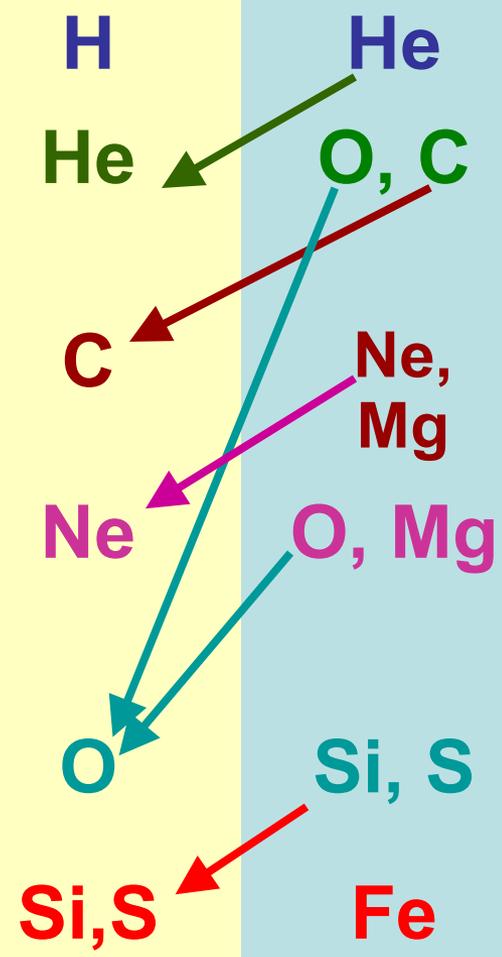
The net result is



Nuclear burning stages

(20 M_⊙ stars)

Fuel	Main Product	Secondary Product	T (10 ⁹ K)	Time (yr)	Main Reaction
H	He	¹⁴ N	0.02	10 ⁷	4 H $\xrightarrow{\text{CNO}}$ ⁴ He
He	O, C	¹⁸ O, ²² Ne s-process	0.2	10 ⁶	3 He ⁴ \rightarrow ¹² C ¹² C(α,γ) ¹⁶ O
C	Ne, Mg	Na	0.8	10 ³	¹² C + ¹² C
Ne	O, Mg	Al, P	1.5	3	²⁰ Ne(γ,α) ¹⁶ O ²⁰ Ne(α,γ) ²⁴ Mg
O	Si, S	Cl, Ar, K, Ca	2.0	0.8	¹⁶ O + ¹⁶ O
Si, S	Fe	Ti, V, Cr, Mn, Co, Ni	3.5	0.02	²⁸ Si(γ,α)...

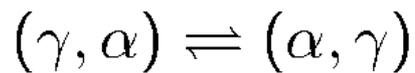


Silicon/Sulfur Burning

Actually, often we have more sulfur in the star than there is silicon, but it is custom to call this phase “silicon burning”.

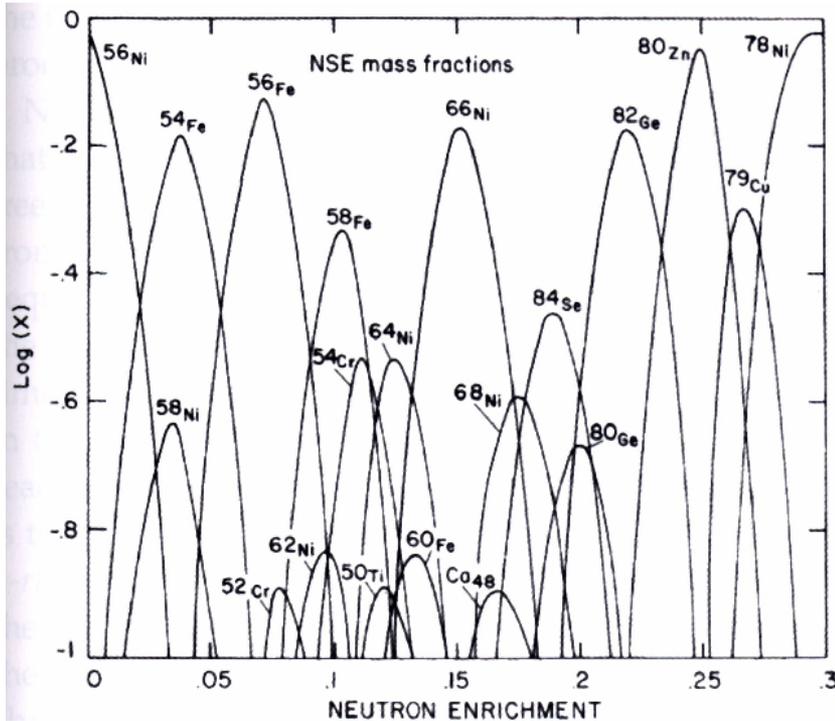
Typical burning temperature is $3 \dots 3.5 \times 10^9$ K.

Similar to neon burning, silicon burning proceeds as a series of photo-disintegration reactions, mostly, (γ, α) , and helium capture reactions, (α, γ) to build up iron group elements.



At the high T and ρ of these conditions, also *weak reactions* occur, converting protons into neutrons and leading to a *neutron excess*. This allows to actually make stable iron isotopes.

Beyond Silicon Burning



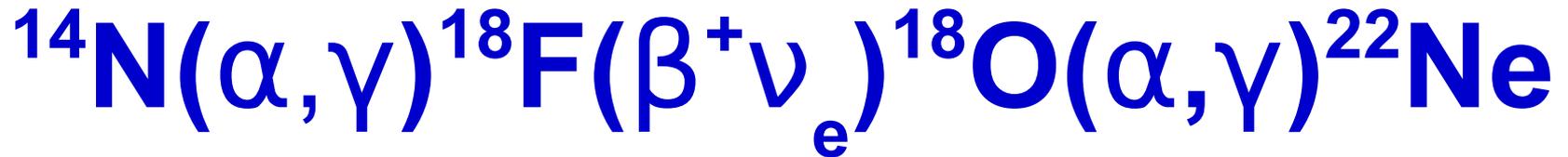
After silicon burning T and ρ is so high that the nuclei are in **nuclear statistical equilibrium**, i.e., the reactions are fast compared to the evolution time-scale of the star, and the abundance distribution of the nuclei is given by a *Saha equation*.

NSE distribution for
 $T = 3.5 \times 10^9 \text{ K}$,
 $\rho = 10^7 \text{ g/cm}^3$

Summary of Energies

<i>Nuclear Fuel</i>	<i>Process</i>	$T_{threshold}$ $10^6 K$	<i>Products</i>	<i>Energy per Nucleon (MeV)</i>
H	$p-p$	~ 4	He	6.55
H	CNO	15	He	6.25
He	3α	100	C, O	0.61
C	$C + C$	600	O, Ne, Na, Mg	0.54
O	$O + O$	1000	Mg, S, P, Si	~ 0.3
Si	Nuc. eq.	3000	Co, Fe, Ni	< 0.18

Nitrogen Burning



- ^{14}N is made as slowest reactant in CNO cycle
- It is made from initial metals, not as a primary product
- Depending on metallicity, the abundance can be come significant; it will be more important for more metal-rich stars.
- ^{14}N burning occurs at the onset – before – central helium burning and can have its own convective burning phase, take a few % of helium burning time.

