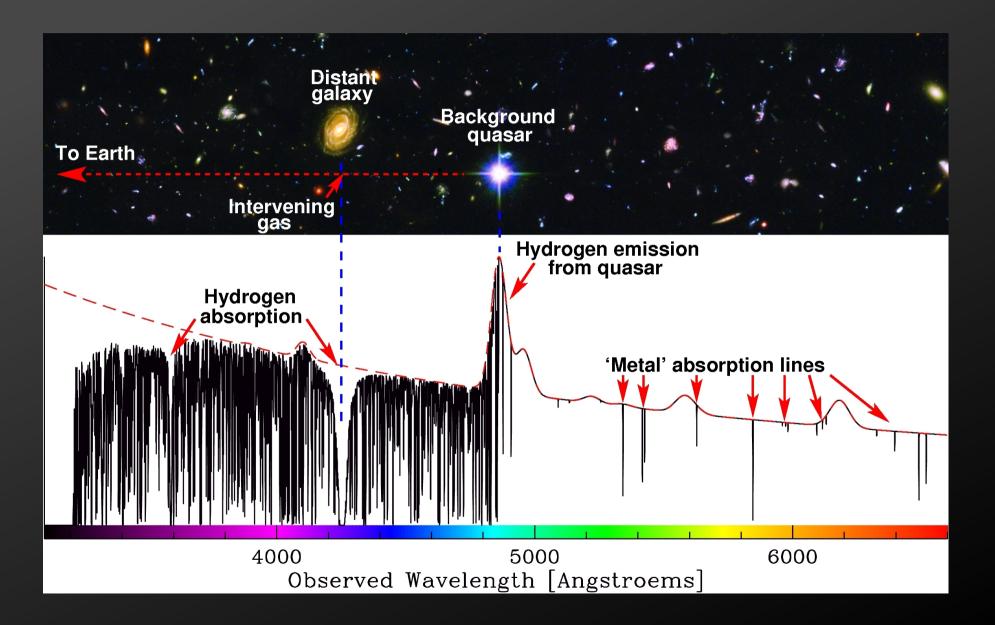
Probing Chemical Evolution with Damped Lyman α Systems

What can DLAs do for you?

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Sara Ellison (Victoria; Supervisor) J.X. Prochaska (UC Santa Cruz) Kim Venn (Victoria)

Quasar absorption line systems

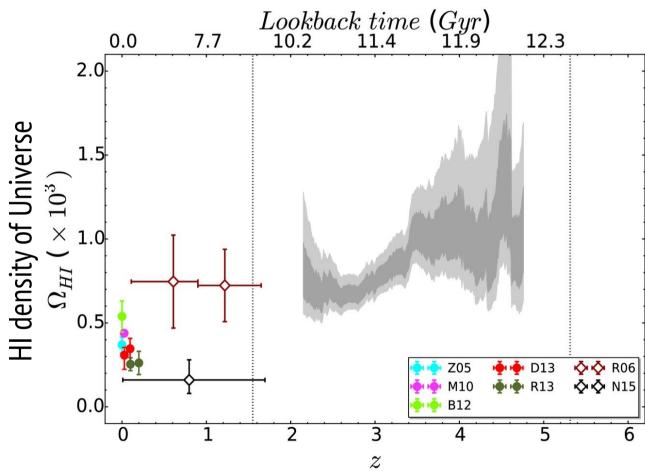


Damped Lyman α Systems (DLAs)

Large HI column
 densities; logN(HI) ≥
 20.3

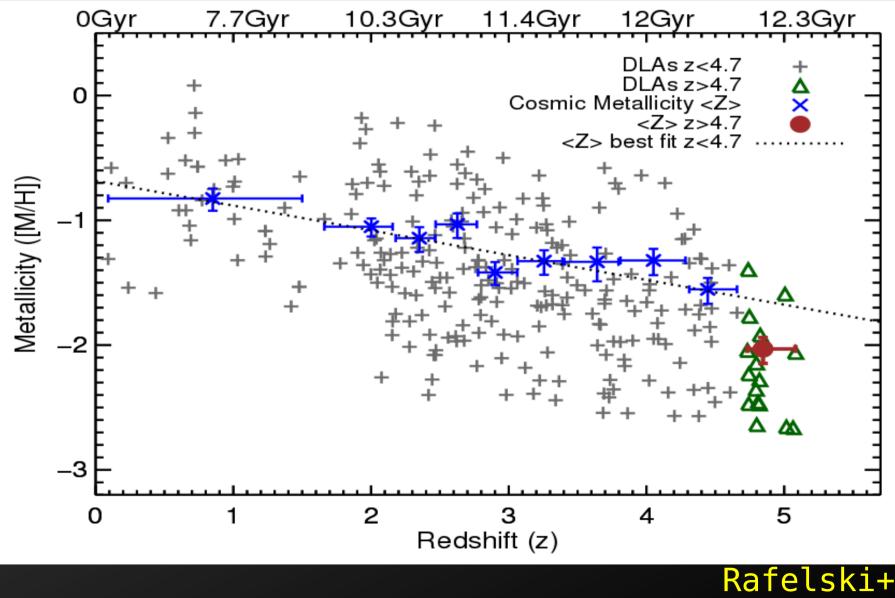
– Galactic ISM columns

• Dominate HI content of Universe



Sanchez-Ramirez+ 2016

Damped Lyman α Systems (DLAs)



2014

Why use DLAs for chemical evolution? Pros Cons • Probe evolution all the way back to z~5

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- Probe evolution all the way back to z~5
- Detailed abundances for systems other than Local Group
- Magnitude and Morphology blind (require gas)

Cons

• Lack of galaxy properties without imaging

Why use DLAs for chemical evolution?

Pros

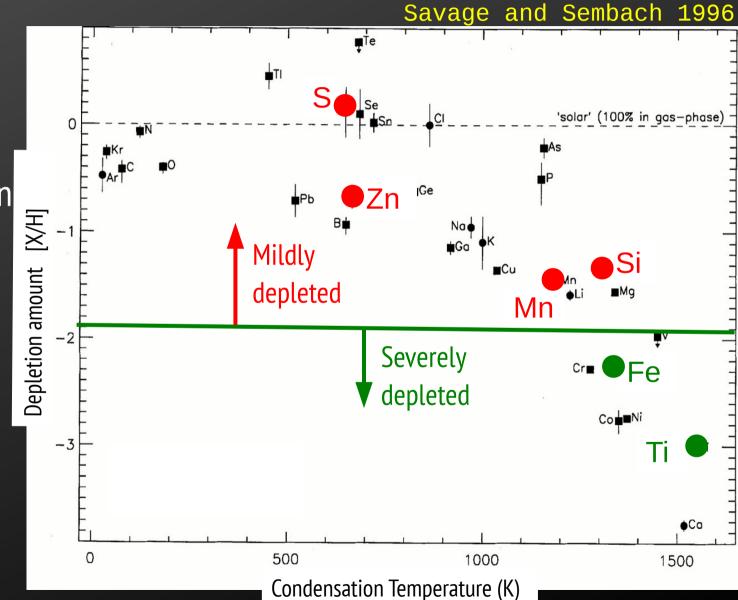
- Probe evolution all the way back to z~5
- Detailed abundances for systems other than Local Group
- Magnitude and Morphology blind (require gas)

Cons

- Lack of galaxy properties without imaging
- Gas phase abundances –
 Dust depletion effects

The problem with DLAs: Dust depletion

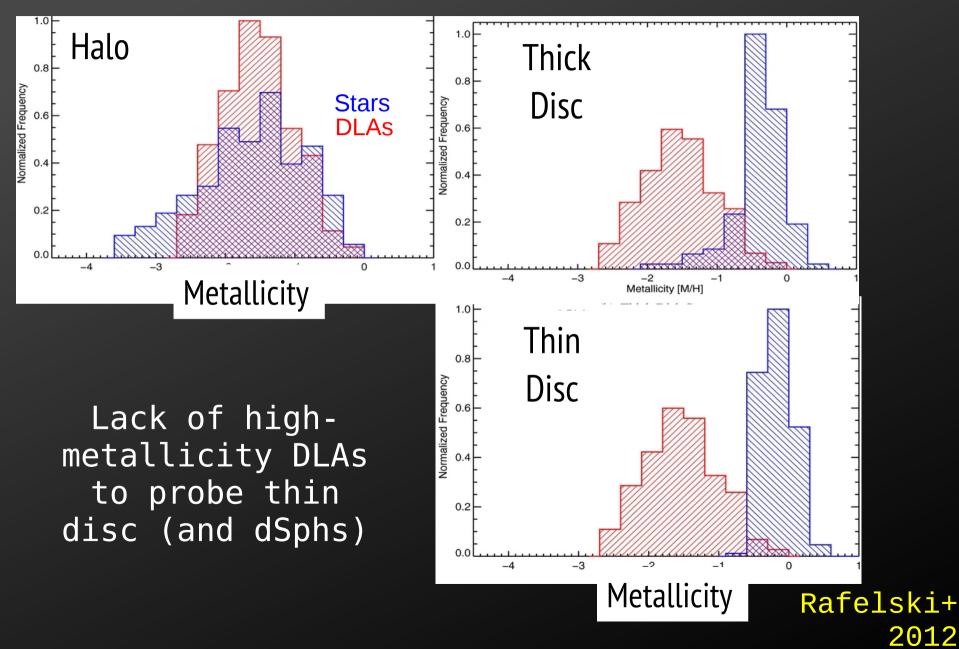
Condensation
 onto dust grains
 –Underprediction
 of total column
 densities



Goal of using abundances of DLAs

Can we use DLA chemistry to understand what environment DLAs arise whilst correcting for dust depletion?

Environments probed by DLAs



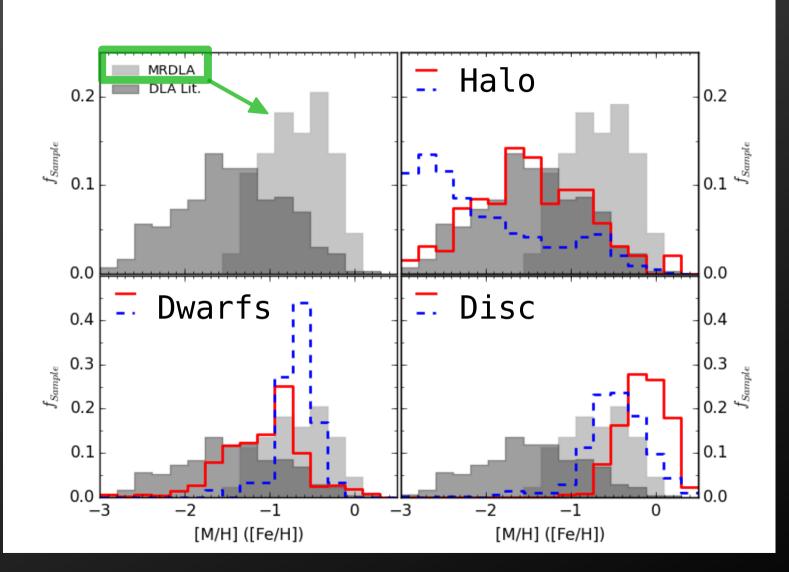
"Metal Rich" DLAs

- Probe galaxies that:
 - Study thin-disk and dSphs environments at z~2 (ISM of stars today?)
 - May contain rarely detected elements (Mn, B, Pb; Prochaska+ 2003)

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- **45 metal-rich DLAs with 100+ hours of Keck/HIRES (Berg et al.** 2015; Paper I)

Environments probed by DLAs



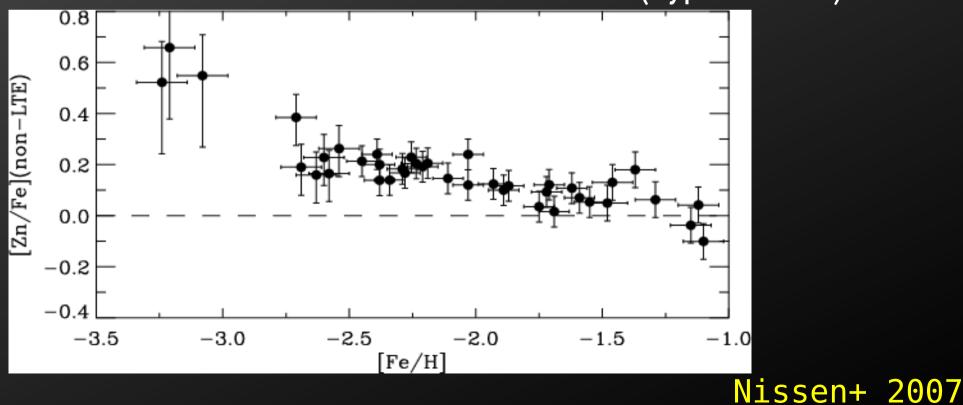
Berg+ 2015 (Paper II)

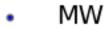
Detailed Chemistry

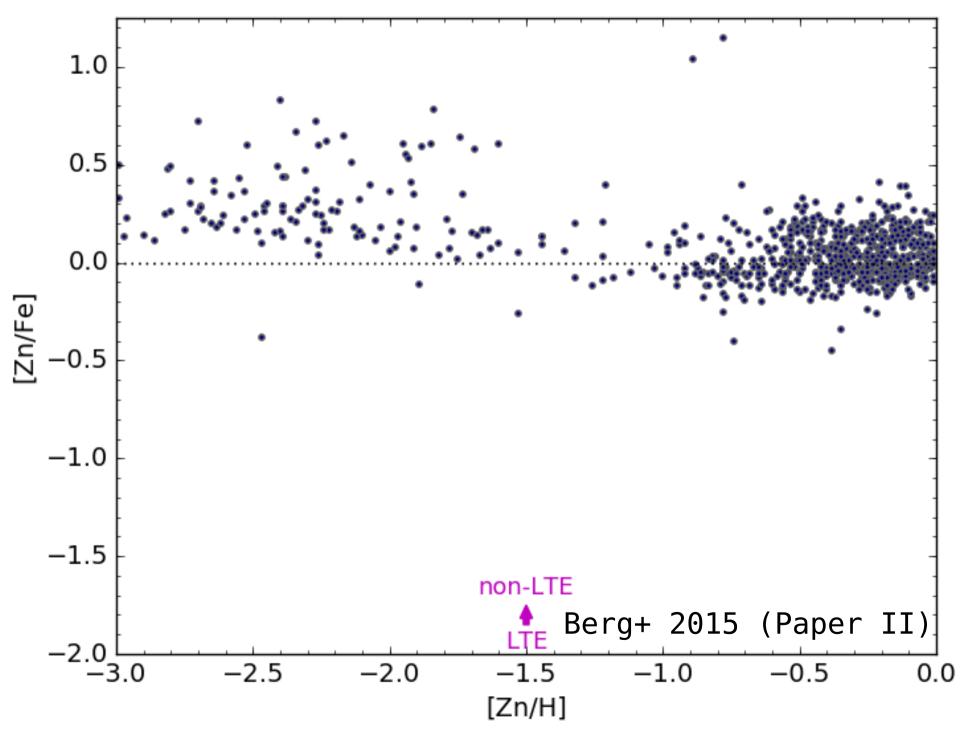
1 1,0,-1 Hydrogen 1,008 3 Li Lithium 6,641 11 1,-1 Nolum 22,090	4 Bee Beryllium 0012 2 12 Magnesium 24.305	2 Jum											6 4321-1 Carbon 14 4321-1 Silicon 3000	7 54.3.2.1.0 Nitrogen 14.007 15 54.3.2.1 15 54.3.2.1 Phosphorus 30.074	8 0 0xygen 16 6.5.4.3.2.1 Sulfur 2006	9 F Fluorine 18.008 17 Chl Chlorine 35.453	2 He Helium Helium 10 Ne Neon 20 180 18 Argon 39.948
19 K Potassium 39.098	20 2 Calcium 40.078	21 3.2. Sc Scandium 44.956	22 -12 Titanium	23 Vanadium	24 0.1.2.3.4 Cr Chromium 51.996	25 ^{7,0,5,4,3,2,} 0,-1,-2,- Manganese 54,938	26 ^{6,5,4,3,2,1} 0,-1,-2 Fe Iron 55.845	27 5.4.3.2.1 01 Cobalt 58.933	28 0.4.3.2.1 01 Nickel 58.693	29 4.3.2.1. Cu Copper 63.546	30 2.1.0 Zn Zinc 65.38	31 ^{3,2,1} Gallium 69.723	32 4.3.2.1 Ge Germanium 72.631	33 5.3.23 Asenic 74.922	34 ^{0.4,2,1,-2} Selenium 78.971	35 7.5.4.3.1.0 Br Bromine 79.904	36 2,0 Krypton 84.798
37 ^{1,-1} Rb Rubidium 84.468	38 2 Sr Strontium 87.62	39 3,2 Y Yttrium 88.906	40 20 21 21 21 21 21 21 21 21 21 21	41 5.4.3.2.1 Nb Niobium 92.906	42 6.5.4.3.2.1 MO Molybdenum 95.95	43 ^{7:03,43,62,1} 0,-1,-3 TC Technetium 98.907	44 1.02 Ru Ruthenium 101.07	45 ^{0.5,4,3,2,1} 0,1 Rhodium 102,906	46 4.2.0 Pd Palladium 106.42	47 3.2.1,0 Ag Silver 107.888	48 2,1 Cd Cadmium 112,414	49 ^{3.2.1} In Indium 114.818	50 4.2.4 Sn Tin 118.711	51 53.3 Sb Antimony 121.760	52 ^{0,5,4,2,-1,-2} Tellurium 127.6	53 7.5.3.1.0.4 Iodine 126.904	54 8.8.4.3.2.0 Xe Xenon 131.294
55 11 Cesium 132.905	56 ² Ba Barium 137.328	57-71	72 4,3,2,1 Hafnium 178,49	73 54.3.2.1 -13 Tantalum 180.948	74 6.5.4.3.2.1 0.12.4 Tungsten 183.84	75 ^{7,6,5,4,3,2,1} Re Rhenium 186,207	76 ^{8,7,6,5,4,3,2} 0,1,0,-2 0,5 0,5 0,5 1,0,-2 1,0,-2	77 ^{6,5,4,3,2,1} 0,-1 Iridium 192,217	78 ^{6,5,4,2,0} Pt Platinum 195.085	79 ^{7,5,3,2,1} 0,-1 AU Gold 196,967	80 2.1 Hg Mercury 200.592	81 3,1 Thallium 204.383	82 4.2 Pb Lead 207.2	83 5.3.13 Bi Bismuth 208.980	84 ^{6,4,2,-2} Polonium [208,982]	85 7.5.3.11 At Astatine 209.987	86 ^{2,0} Radon 222.018
87 1 Francium 223.020	88 Ra Radium 226.025	89-103	104 4.3 Ref Rutherfordium	105 5.4 Db Dubnium [262]	106 ^{6,5,4} Sg Seaborgium [266]	107 ^{7,6,5,4,3} Bh Bohrium [264]	108 4.3.2.1 HS Hassium [269]	109 ^{6,5,4,3,2,1} Mt Meitnerium [268]	110 ^{6,5,4,3,2,1} DS Darmstadtium [269]	111 31 Rg Roentgenium [272]	112 ^{2,1} Cn Copernicium [277]	113 Uut Ununtrium unknown	114 ² Flerovium [289]	115 ^{3,1} Uup Ununpentium unknown	116 ^{4.2} Lv Livermorium [298]	117 ^{unknown} Uus Ununseptium unknown	118 8.6.4.2 Uuo Ununoctium unknown

[Zn/Fe] – dust tracer or nucleosynthetic probe?

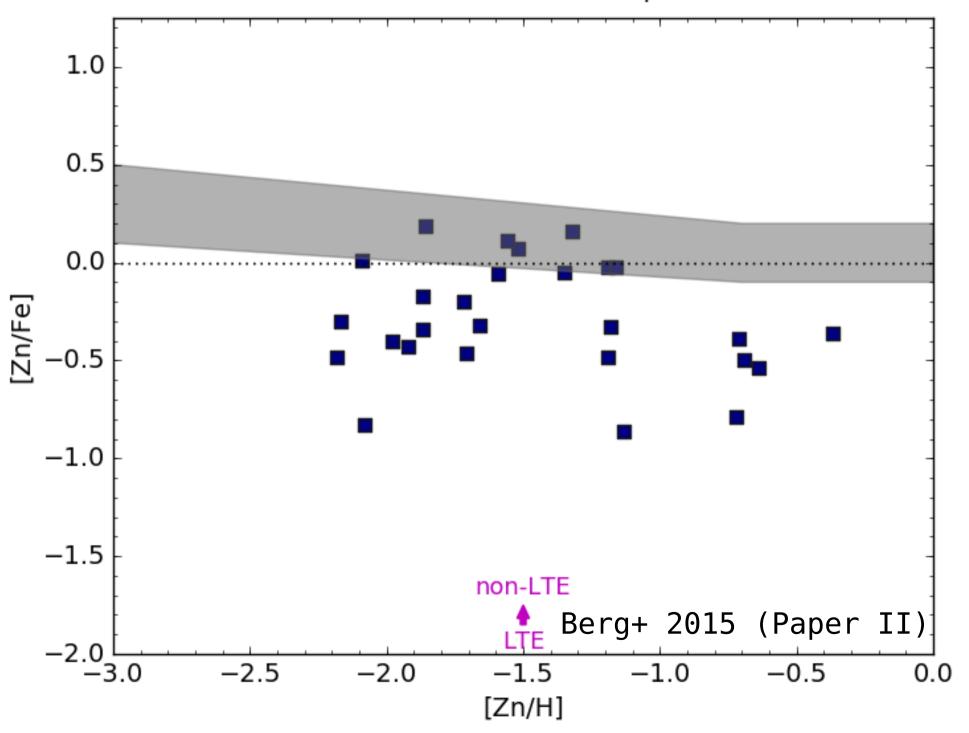
- Forms in lockstep with Fe at solar metallicities
 - DLA dust tracer (e.g. Pettini+ 1994)?
- Different contribution at low metallicities (hypernovae?)

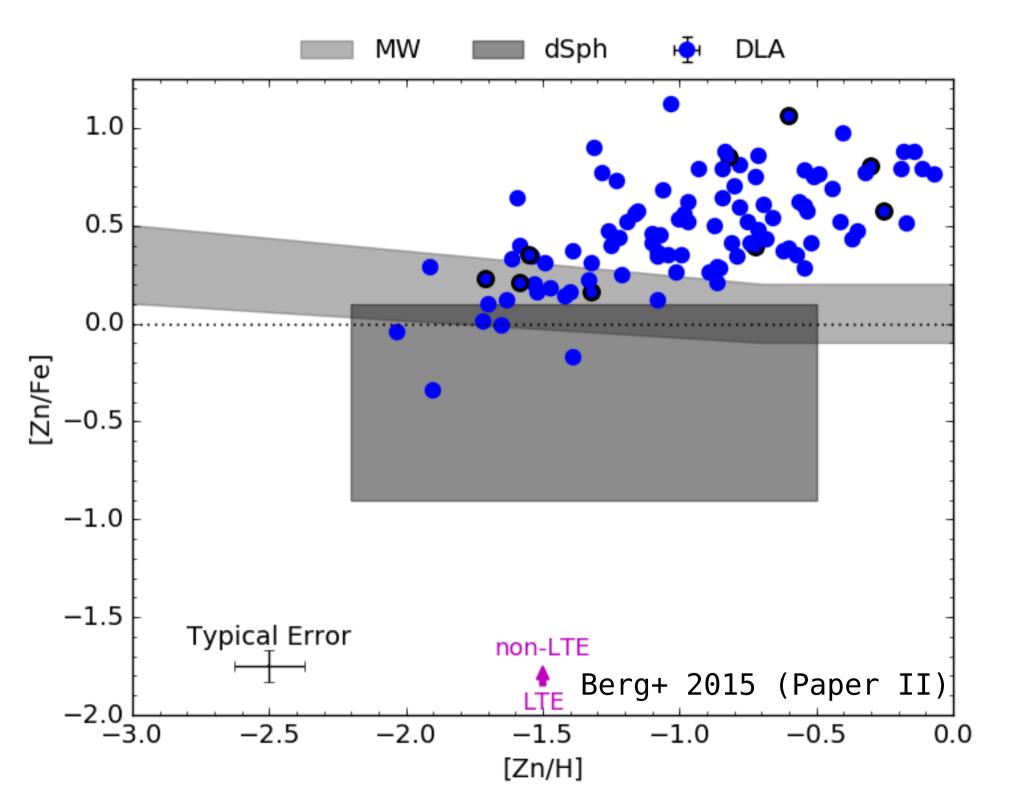






🛛 MW 🧧 dSph

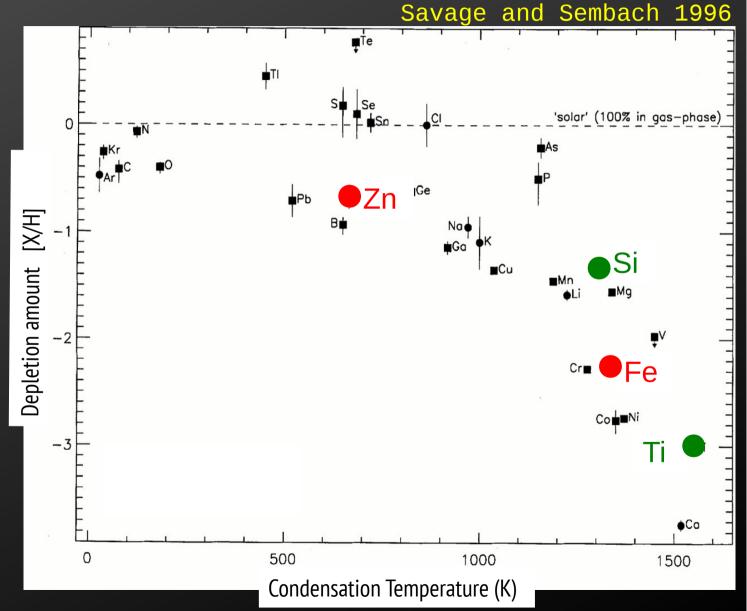




The problem with DLAs: Dust depletion

Relative
 depletion patterns
 should scale from
 galaxy to galaxy

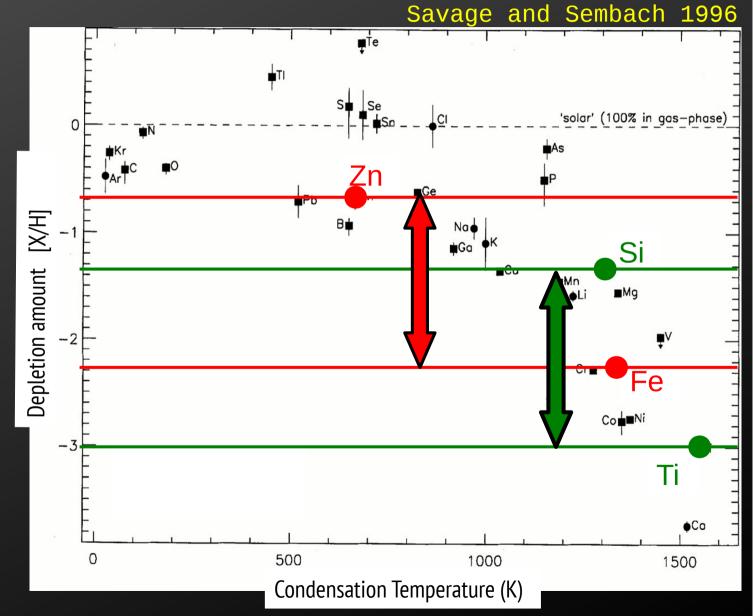
Elements with
similar
nucleosynthetic
origin should give
dust scaling

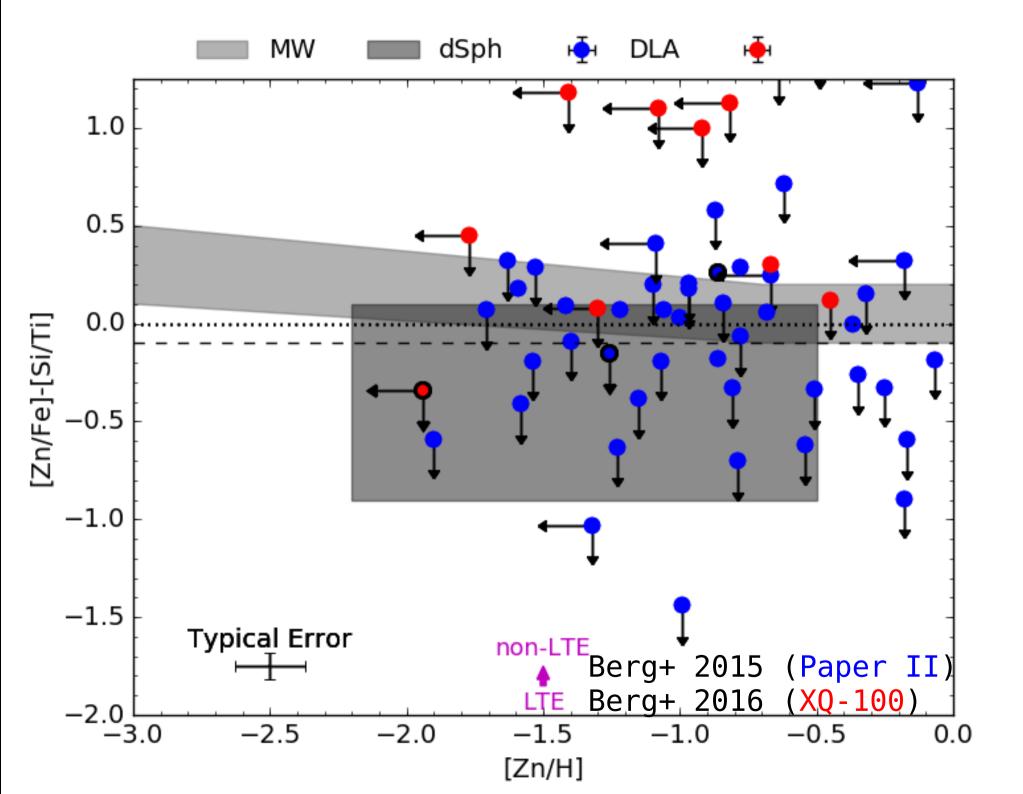


The problem with DLAs: Dust depletion

Si and Ti both
 alpha-elements;
 same relative
 depletion as Zn-Fe

• [Zn/Fe] – [Si/Ti] will give the nucleosynthetic contribution of [Zn/Fe]





Zn an Fe-peak tracer?

- Zn is not an Fe-peak tracer for ALL galaxies
- Some DLAs are similar to dSph [Zn/Fe] values

-Others are consistent with both MW and dwarfs

• Need *accurate* Zn abundances in different environments!

DLA chemistry summary

- DLAs offer probe of observing chemical evolution across cosmic time
- MRDLAs can probe regime where differences in chemistry are seen in Local Group environments
 - [Zn/Fe] Consistent with MW and dSphs
 - $[\alpha/Zn] Low \alpha$ in DLAs
 - [Mn/Fe] Like dSphs (at high metallicities)
- Require handle on odd-Z and Zn in various environments