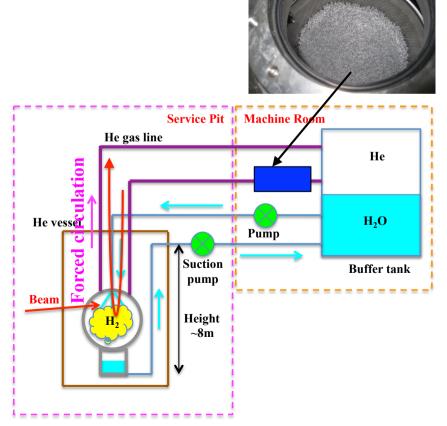
# Water Radiolysis Study for Hydrogen Recombination System

T. Sekiguchi (KEK)

2019/3/20

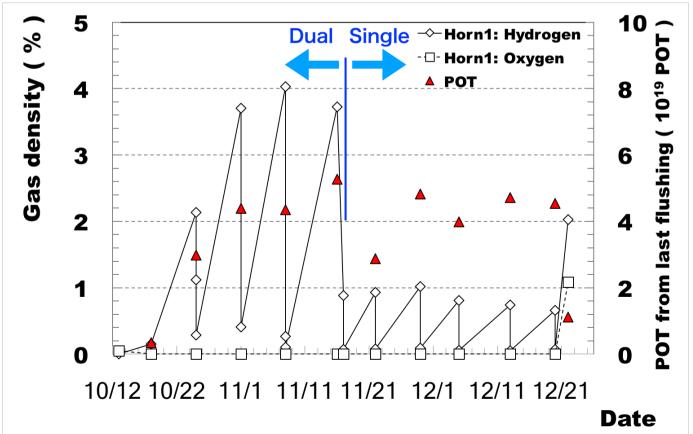
## H<sub>2</sub> Production Issue

- H<sub>2</sub> production by water radiolysis
  - Final products: H<sub>2</sub>, H<sub>2</sub>O<sub>2</sub>, O<sub>2</sub> (=1.3:1:0.1)
  - Measured production rate @ 750 kW
    - 260 L (5%) /10<sup>19</sup> POT (~1day)
- H<sub>2</sub> recombination (2H<sub>2</sub> + O<sub>2</sub>  $\rightarrow$  2H<sub>2</sub>O)
  - Recombination by a catalyst (alumina pellet w/ 0.5% Pd)
  - Forced He flow inside the horns
    - Recombination should work efficiently
- Status
  - Can keep H<sub>2</sub> concentration below 1%
    - Weekly He flushing 1% → 0.1%



# Effect of Ion Exchanger

- From fall 2017
  - Dual ion-exchangers (IE) are used  $\Rightarrow$  H<sub>2</sub> concentration increased to 4% level.
- Back to the single IE operation
  - H<sub>2</sub> concentration was reduced by 1/2~1/4
  - IEs cause the higher H<sub>2</sub> concentration.



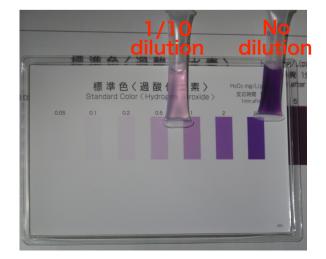
IE may be damaged by H<sub>2</sub>O<sub>2</sub>

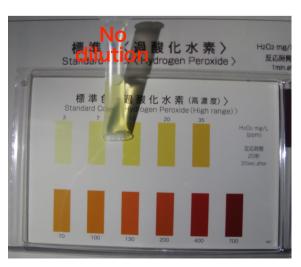


Is H<sub>2</sub> produced due to oxidization of resins by H<sub>2</sub>O<sub>2</sub>?

# Effect of H<sub>2</sub>O<sub>2</sub>

- Hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>)
  - Production rate is close to that of H<sub>2</sub> (H<sub>2</sub>: H<sub>2</sub>O<sub>2</sub> = 1.3: 1)
  - It corrodes IE resins ⇒ degradation of IE with presence of H<sub>2</sub>O<sub>2</sub>
  - Can be decomposed naturally: 2H<sub>2</sub>O<sub>2</sub> ⇒ 2H<sub>2</sub>O + O<sub>2</sub>
    - Catalyst can accelerate the decomposition
    - Resultant O<sub>2</sub> can work as a source O<sub>2</sub> for the recombination
- Measurement of H<sub>2</sub>O<sub>2</sub> concentration
  - ~10 mg/L ⇔ estimated : ~500 mg/L @ 3.9x10<sup>20</sup> POT
    - This indicates that most of H<sub>2</sub>O<sub>2</sub> produced was decomposed
  - Even this small concentration of H<sub>2</sub>O<sub>2</sub> can affect the IE resins

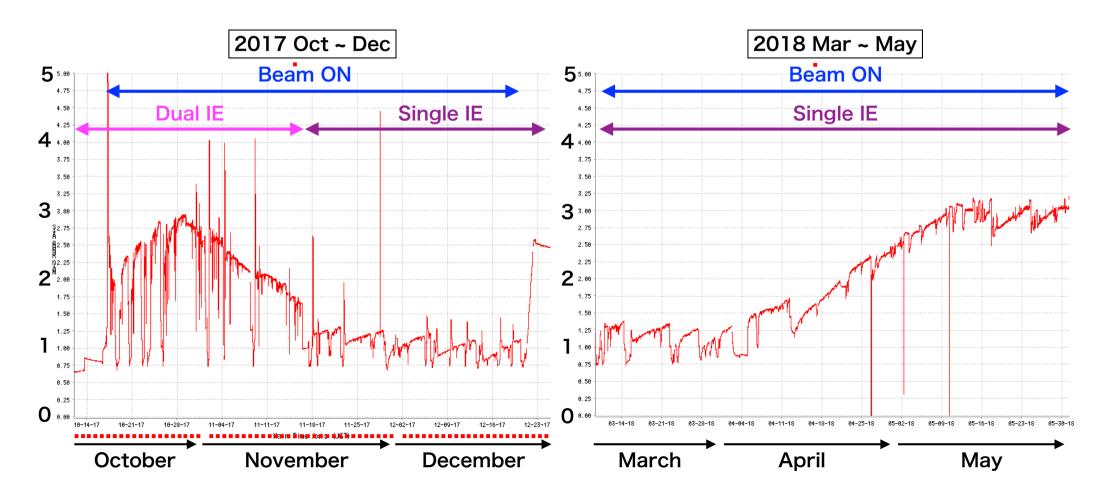




## Water Conductivity Trend

## Water conductivity

- A strange behavior during dual IE operation
- Low conductivity during single IE operation
- Conductivity got increased around middle of April
  - This indicates lifetime of IE  $\Rightarrow$  ~6.0 x 10<sup>20</sup> POT



## H<sub>2</sub>O<sub>2</sub> Resistant Ion Exchanger

- H<sub>2</sub>O<sub>2</sub> is problematic for nuclear reactor business
  - H<sub>2</sub>O<sub>2</sub> resistant ion exchanger developed for reactor
  - Pd-doped ion exchanger can decompose H<sub>2</sub>O<sub>2</sub> and ⇒ Pd works as a catalyst
    - therefore can extend lifetime of ion exchanger

**Award Paper** 

T. Izumi et al (2018)

#### Prolongation Technology of Life Time of Ion Exchange Resins in Nuclear Power Plants

Takeshi IZUMI  $^{1\ast}$ , Makoto KOMATSU  $^{1}$ , Tatsuya DEGUCHI  $^{2}$ 

<sup>1</sup>EBARA Corporation, 4-2-1 Honfujisawa, Fujisawa-shi, Kanagawa 251-8502, Japan <sup>2</sup>EBARA Corporation, 11-1 Haneda Asahi-cho, Ota-ku, Tokyo 144-8510, Japan

(Manuscript accepted July 20, 2018)

#### **Abstract**

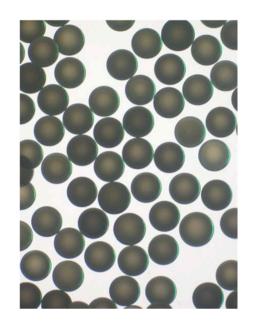
From the viewpoint of the minimizing the corrosion of the reactor's structural material in the nuclear power plants, the ion exchange resins are generally used as one of purification system in order to keep water quality clean. Hydrogen peroxide generated by the radiolysis of water exists in the reactor water and it accelerates the oxidation decomposition of the ion exchange resins and finally, it becomes the cause to shorten the resin life. To solve this problem, the application of Pd doped resins which can decompose hydrogen peroxide catalytically at the surface has been considered. It was confirmed by the cold test that Pd doped resins overlaid on the ion exchange resins or mixed with the ion exchange resins decomposed hydrogen peroxide contained in the reactor water and inhibited the oxidative degradation of the ion exchange resins. We report the results of these tests.

Keywords: Pd doped resins, Hydrogen peroxide, Ion exchange resins, Nuclear power plants

## Status

### New resin

- I contacted the developer
- He was very interested in the application of this resin
- As a trial, this resin was put into one IE bottle for a test
  - 4L of the old resins were replaced with new ones
- To be used during next beam time



Remove old resins



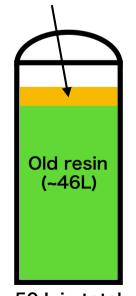
Put new resins



**Completed IE bottle** 



New resin (~4L) is overlaid



50 L in total

# Summary

## Water radiolysis

- Final products are H<sub>2</sub>, H<sub>2</sub>O<sub>2</sub>, O<sub>2</sub>
- H<sub>2</sub> production rate @ 485kW : 260 L (5%) /10<sup>19</sup> POT
- $H_2$  recombination to remove  $H_2 \Rightarrow 1\%$  / week
- Ion exchanger condition affects H<sub>2</sub> production rate
- H<sub>2</sub>O<sub>2</sub> affect the performance of ion exchangers

## • H<sub>2</sub>O<sub>2</sub>

- Degradation of ion exchanger lifetime was observed around 6x10<sup>20</sup> POT
  - Oxidization of the resins by H<sub>2</sub>O<sub>2</sub>
- Pd-overlaid ion exchanger resins can work as a catalyst for H<sub>2</sub>O<sub>2</sub> decomposition
- lon exchanger bottle with new resins will be tested during next beam time to check its performance