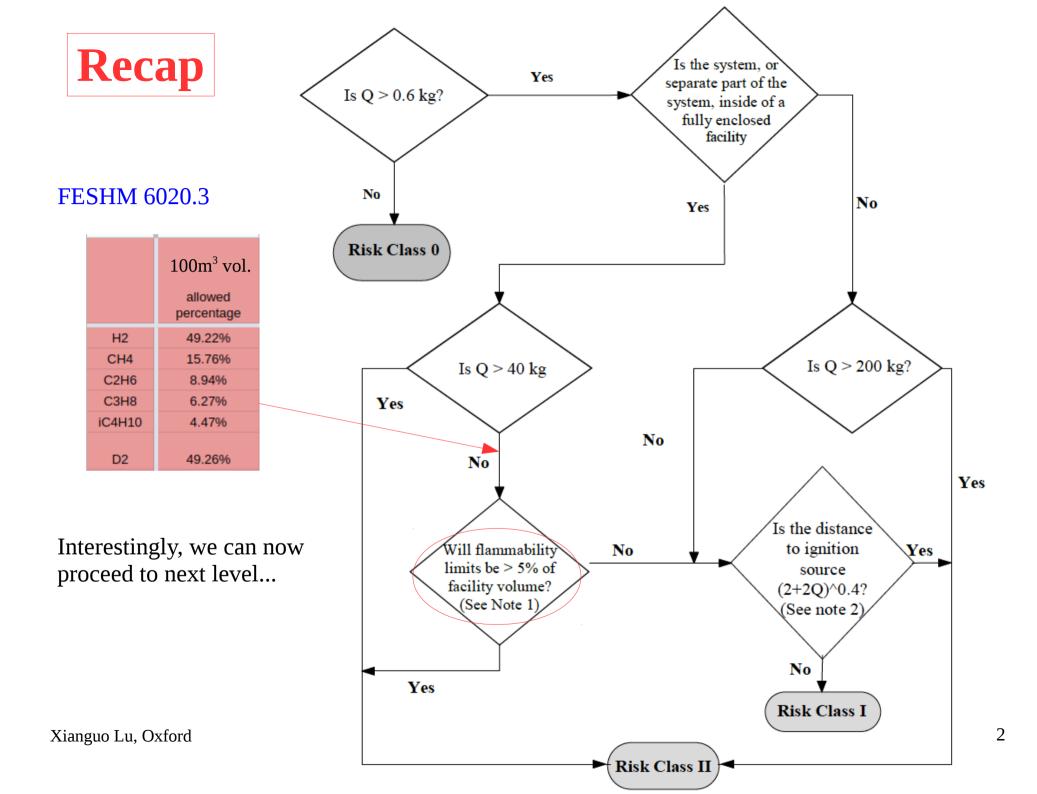
Gas search for neutrino-hydrogen interactions with DUNE HPgTPC – updated version (see UPDATE tag)

Xianguo Lu University of Oxford

DUNE-ND Gas Meeting 2020 June 2

*with input from Diego Diaz and Philip Hamacher-Baumann



Example 1

Two 81 SCF cylinders of a 50-50 mixture (by volume) of argon-ethane (Fermilab stock catalog number 1980-1095) will be used in a room whose volume is 9*15*20 ft³ (2700 ft³). This room. inside a larger building, contains no obvious fire hazards such as welding operations. The gas is to be supplied to drift chambers. First, to determine Q, it is recognized that only 40.5 SCF of a given cylinder is ethane. Thus, from Appendix 3 and Appendix 4;

 $O = 2*40.5 \text{ ft}^3 * 0.028 (m^3/\text{ft}^3) * 1.26 (kg/m^3) * 0.36 (H_2 \text{ equivalence factor})$

Q = 1.03 kg hydrogen equivalent inventory

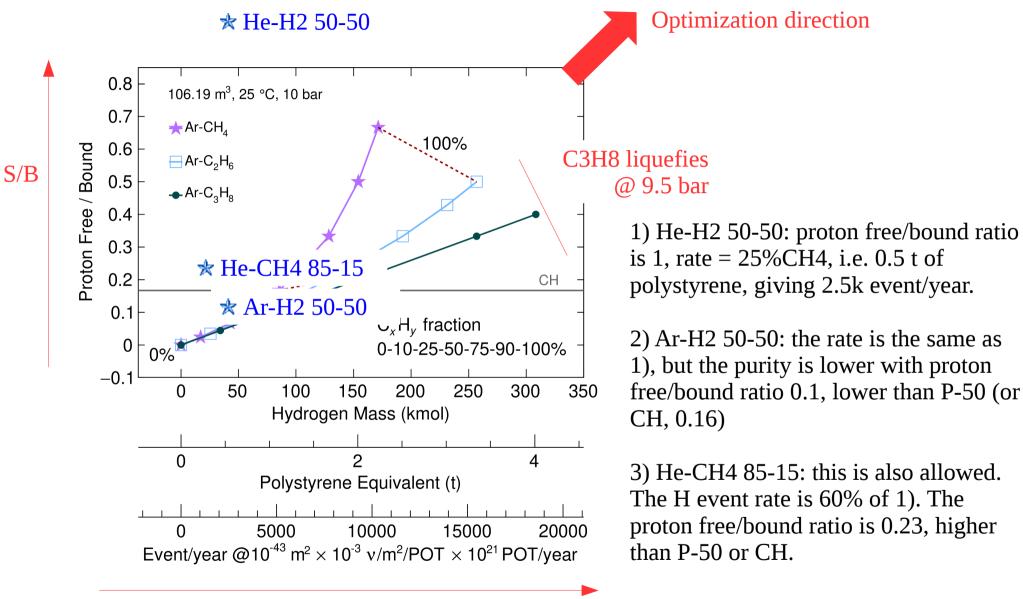
Thus by box 1 in the flowchart, we exceed the limit for Risk Class 0 and must go to box 2. Continuing to box 2, we find the answer to be yes but the answer to the question in box 3 is negative. Doing the calculation prescribed in box 4 we find that 5% of 2700 ft³ is 135 ft³. Dividing 81/135 finds a maximum concentration of 60 %, which exceeds the flammability upper limit. Thus, any concentration below this limit is reachable with the available inventory, since no inventory controls have been specified. Therefore the answer to this question is affirmative and the Risk Class is II. If only a single cylinder was needed, the Ø.5 kg hydrogen equivalence would have rendered a Risk Class 0 determination.

Recap	UPDATE
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FESHM 6020.3 example

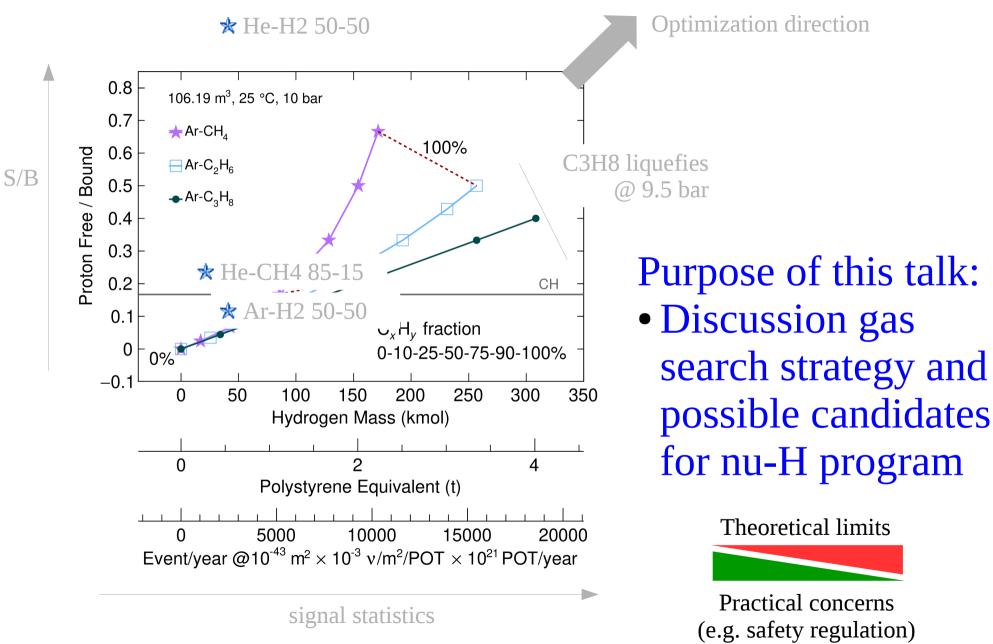
o determination	•		
	flammability upper limit in % (from FESHM 6020.3 p17)	minimal facility volume (m^3) = gas volume at 1 bar/upper flam limit/5%	Minimal facility volume at maximal allowed mass
H2	75	1.31E+04	Just match the current hall size 1.3E4 m^3
CH4	15	2.10E+04	Very large due to the small upper limit
C2H6	12.5	1.43E+04	
C3H8	9.5	1.32E+04	
D2	75	1.31E+04	
			3





4

Introduce **H-index** = f/b * #H

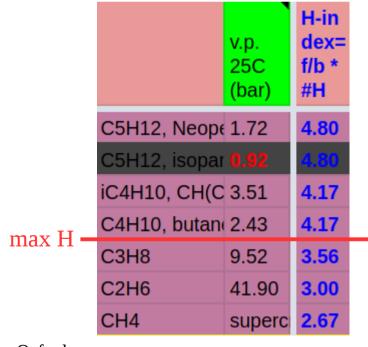


Maximal hydrogen event rate with alkane

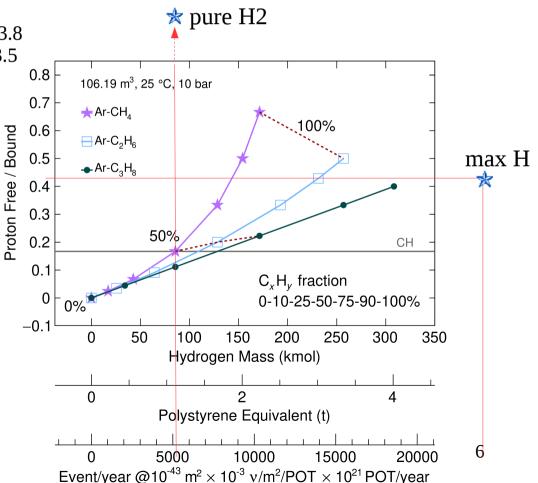
- CnH2n+2: most efficient hydrogen carrier
 - \checkmark Large n → high H event rate
 - ^{*} Large $n \rightarrow low$ vapor pressure
 - from propane (C3) on can't get to 10bar, last gas at 1bar is neopentane (C5)
- **Maximal H** achieved by towering partial pressure in 10bar :

17% neopentane (C5) + 35% isobutane (C4) + (<u>1-17%-35%=</u>) 48% propane (C3) = C3.7H9.4

- <u>f/b: 0.42, H-index: 4.0</u>
- 1) 17% neopentane (C5) + 35% isobutane (C4) + 24% butane (C4) + (1-17%-35%-24%=) 24% propane (C3) = C3.9H9.9
 - f/b: 0.42, H-index: 4.1
- 2) Other "towering":
 - > 35% C4 + 65% C3 = C3.4H8.7, f/b: 0.43, H-index: 3.8
 - > 95% C3 + 5% C2 = C3.0H7.9, f/b: 0.44, H-index: 3.5
- Corresponding maximal event rate: ~25k/year

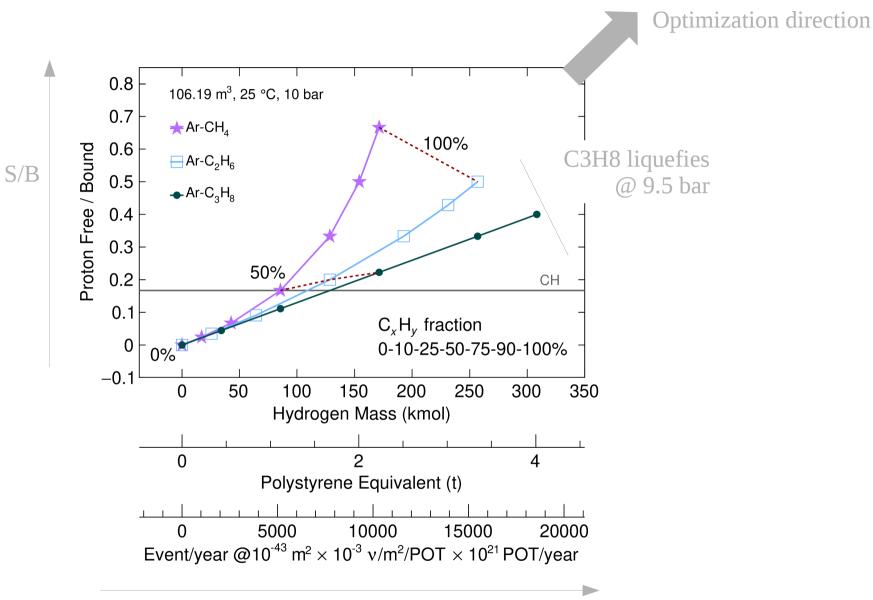


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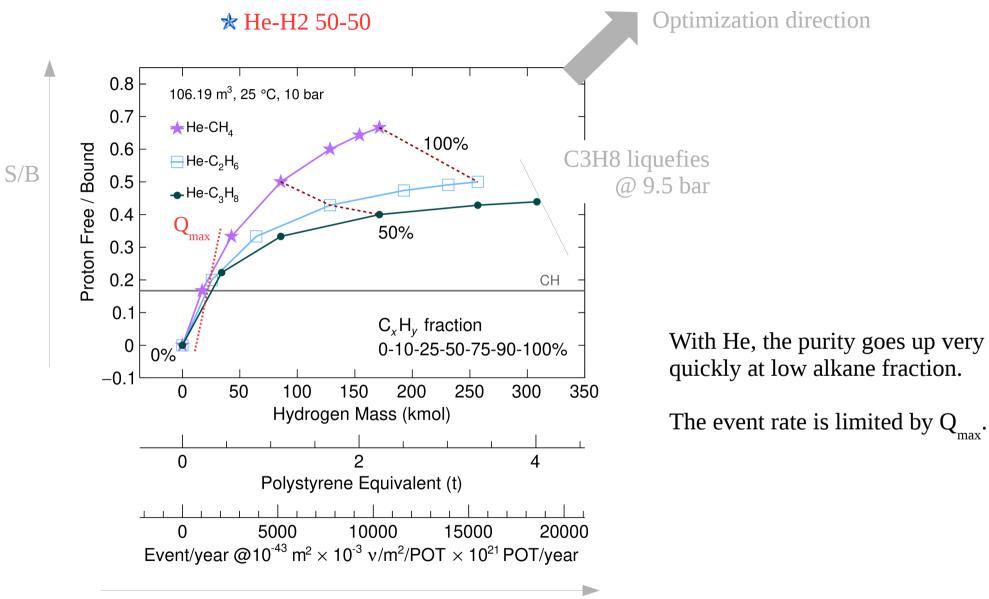
UPDATE





signal statistics

He-base mixtures



signal statistics

Search for H-rich non-flammable gas

Safety Classification of	Refrigerants ^[47]				
Flammability in Air @ 60 °C & 101.3 kPa	ASHRAE 34 Safety group				
Higher Flammability LFL or ETFL ₆₀ = 100 g/m ³ OR HOC = 19 MJ/kg ^[48]	A3	В3			
Lower Flammability LFL or ETFL ₆₀ > 100 g/m ³ & HOC < 19 MJ/kg ^[48]	A2	B2			
Lower Flammability LFL or ETFL ₆₀ > 100 g/m ³ & HOC < 19 MJ/kg ^[48] with a maximum burning velocity of = 10 cm/s	A2L	B2L			
No flame Propagation	A1	B1			
Flammability in Air @ 60 °C & 101.3 kPa	Lower Toxicity OEL ≥ 400 ppm ^[25]	Higher Toxicity OEL < 400 ppm ^[25]			

ASHRAE 34 Safety group

(ASH-ray) American Society of Heating, Refrigerating and Air-Conditioning Engineers

A3: H2 and all alkane (all LFL <= 5%, see later slides)

LFL is an important parameter in safety, but not considered in FESHM 6020.3

In mixture

x% flammable A + (1-x%) non-fl. B
if x < LFL, the mixture is non-flammable
(assuming A and B stay together if leaked,
namely both have similar density)</pre>

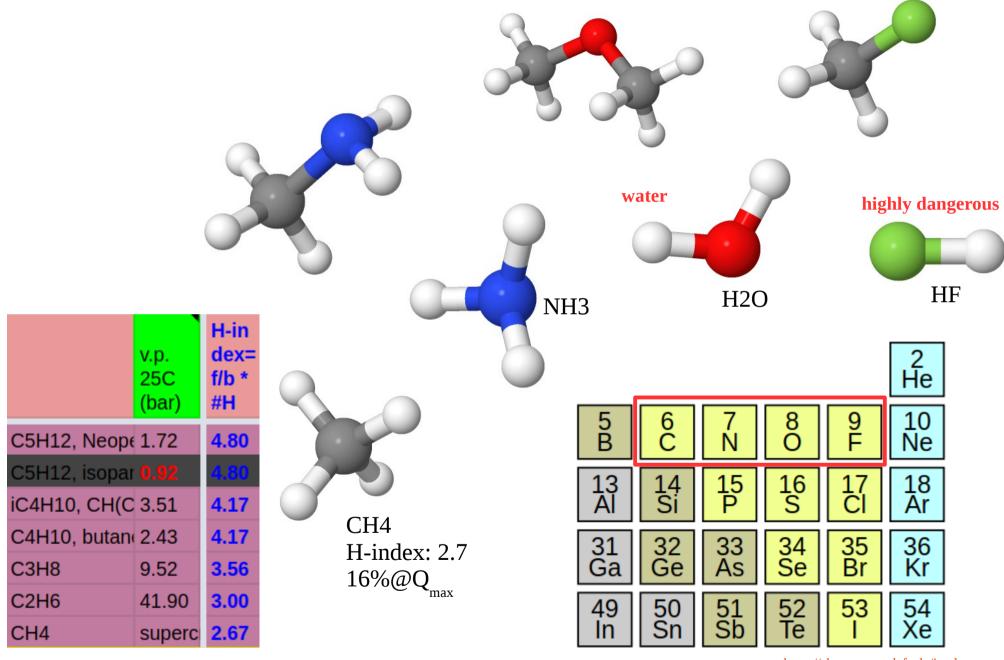
LFL = Lower Flammability Limit

ETFL₆₀ = Elevated Temperature Flame Limit @ 60 °C

HOC = Heat Of Combustion

https://en.wikipedia.org/wiki/List_of_refrigerants

Search for H-rich non-flammable gas



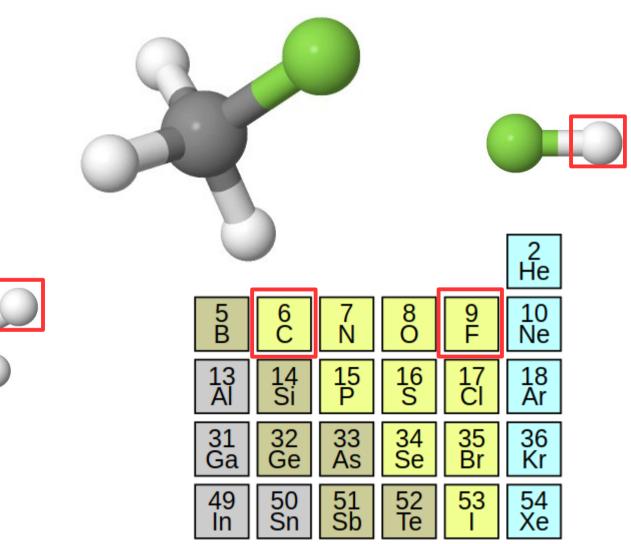
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Check the industry of refrigerants https://en.wikipedia.org/wiki/List_of_refrigerants

C3HF7 1,1,1,2,3,3,3-Heptafluoropropane

CH3F Fluoromethane



https://chemapps.stolaf.edu/jmol

			-1	nax 🔻					
	code name: HFC-, R-		v.p. 25C (bar)	allowed percent age	ASHRA E 34 Safety group	flamma blility lower limit %	fraction of Q at LFL	flammab ility upper limit in	H-in dex= f/b * #H
		H2	superc	49.2%	A3	4	8%	75	999.0
. (C5H12, Neopentane, C(CH3)4, 2,2-Dimethy	1.72	3.8%	A3?	<u>1.4</u>	36%	7.5	4.80
		C5H12, isopantane, (CH3)2CHCH2CH3	0.92	0.0%	A3	<u>1.4</u>	7376%	8.3	4.80
	600a	iC4H10, CH(CH3)2CH3	3.51	4.5%	A3	<u>1.8</u>	40%	8.4	4.17
CH <		C4H10, butane	2.43	4.7%	A3	<u>1.8</u>	38%	8.5	4.17
		C3H8	9.52	6.3%	A3	2.1	33%	9.5	3.56
		C2H6	41.90	8.9%	A3	3	34%	12.5	3.00
		CH4	superc	15.8%	A3	5	32%	15	2.67
	152a	C2H4F2 1,1-Difluoroethane	5.96	12.8%	A2	<u>3.7</u>	29%	18	0.53
	143a	C2H3F3 1,1,1-Trifluoroethane	12.62	16.0%	A2L	7	44%	<u>17</u>	0.23
	32	CH2F2 Difluoromethane	17.31	28.4%	A2L	<u>13</u>	46%	<u>33</u>	0.17
	245fa	C3H3F5, 1,1,1,3,3-Pentafluoropropane	1.48		B1				0.14
	134a	C2H2F4 1,1,1,2-Tetrafluoroethane, Norflurar	6.65		A1			x	0.08
HFC*	1234ze	C3H2F4, 1,3,3,3-Tetrafluoropropene	7.00	#DIV/0!	<u>A2L</u>	х	#DIV/0!	х	0.07
	1234yf	C3H2F4 2,3,3,3-Tetrafluoropropene	0.06	#VALUE	A2L		#VALUE		0.07
	236fa	C3H2F6, 1,1,1,3,3,3-Hexafluoropropane	2.72		A1				0.06
	23	CHF3 Fluoroform	45.98		A1			35.3	0.03
	125	C2HF5 Pentafluoroethane	13.90		A1			x	0.02
	227ea	C3HF7, 1,1,1,2,3,3,3-Heptafluoropropane	4.55		A1				0.01

*less common ones in BACKUP

A1, B1: non-flammable A2(L): low flammability

	code name: HFC-, R-		v.p. 25C (bar)	allowed percent age	ASHRA E 34 Safety group	flamma blility lower limit %	fraction of Q at LFL	flammab ility upper limit in	H-in dex= f/b * #H
		H2	superc	49.2%	A3	4	8%	75	999.0
		C5H12, Neopentane, C(CH3)4, 2,2-Dimethy	1.72	3.8%	A3?	<u>1.4</u>	36%	7.5	4.80
		C5H12, isopantane, (CH3)2CHCH2CH3	0.92	0.0%	A3	<u>1.4</u>	7376%	8.3	4.80
	600a	iC4H10, CH(CH3)2CH3	3.51	4.5%	A3	<u>1.8</u>	40%	8.4	4.17
CH <		C4H10, butane	2.43	4.7%	A3	<u>1.8</u>	38%	8.5	4.17
		C3H8	9.52	6.3%	A3	2.1	33%	9.5	3.56
		C2H6	41.90	8.9%	A3	3	34%	12.5	3.00
		CH4	superc	15.8%	A3	5	32%	15	2.67
. (152a	C2H4F2 1,1-Difluoroethane	5.96	12.8%	A2	<u>3.7</u>	29%	18	0.53
	143a	C2H3F3 1,1,1-Trifluoroethane	12.62	16.0%	A2L	7	44%	<u>17</u>	0.23
	32	CH2F2 Difluoromethane	17.31	28.4%	A2L	<u>13</u>	46%	<u>33</u>	0.17
	245fa	C3H3F5, 1,1,1,3,3-Pentafluoropropane	1.48		B1				0.14
	134a	C2H2F4 1,1,1,2-Tetrafluoroethane, Norflurar	6.65		A1			x	0.08
HFC*	1234ze	C3H2F4, 1,3,3,3-Tetrafluoropropene	7.00	#DIV/0!	<u>A2L</u>	х	#DIV/0!	х	0.07
	1234yf	C3H2F4 2,3,3,3-Tetrafluoropropene	0.06	#VALUE	A2L		#VALUE	!	0.07
	236fa	C3H2F6, 1,1,1,3,3,3-Hexafluoropropane	2.72		A1				0.06
	23	CHF3 Fluoroform	45.98		A1			35.3	0.03
	125	C2HF5 Pentafluoroethane	13.90		A1			x	0.02
	227ea	C3HF7, 1,1,1,2,3,3,3-Heptafluoropropane	4.55		A1				0.01

*less common ones in BACKUP

A1, B1: non-flammable A2(L): low flammability

Sorted by H-index

						5010	ubyi	I-mucz		
	code name: HFC-, R-		v.p. 25C (bar)	allowed percent age	ASHRA E 34 Safety group	flamma blility lower limit %	fraction of Q at LFL	flammab ility upper limit in	H-in dex= f/b * #H	
		H2	superc	49.2%	A3	4	8%	75	999.0	
		C5H12, Neopentane, C(CH3)4, 2,2-Dimethy	1.72	3.8%	A3?	<u>1.4</u>	36%	7.5	4.80	Flammabili
		C5H12, isopantane, (CH3)2CHCH2CH3	0.92	0.0%			7376%	8.3	4.80	~ H-index
	600a	iC4H10, CH(CH3)2CH3	3.51	4.5%	A3	<u>1.8</u>	40%	8.4	4.17	
CH		C4H10, butane	2.43	4.7%	A3	<u>1.8</u>	38%	8.5	4.17	
		C3H8	9.52	6.3%	A3	2.1	33%	9.5	3.56	T
		C2H6	41.90	8.9%	A3	3	34%	12.5	3.00	
		CH4	superc	15.8%	A3	5	32%	15	2.67	
. (<mark>152a</mark>	C2H4F2 1,1-Difluoroethane	5.96	12.8%	A2	<u>3.7</u>	29%	18	0.53	
	143a	C2H3F3 1,1,1-Trifluoroethane	12.62	16.0%	A2L	7	44%	<u>17</u>	0.23	
	32	CH2F2 Difluoromethane	17.31	28.4%	A2L	<u>13</u>	46%	<u>33</u>	0.17	
	245fa	C3H3F5, 1,1,1,3,3-Pentafluoropropane	1.48		B1				0.14	
	134a	C2H2F4 1,1,1,2-Tetrafluoroethane, Norflurar	6.65		A1			x	0.08	
	1234ze	C3H2F4, 1,3,3,3-Tetrafluoropropene	7.00	#DIV/0!	<u>A2L</u>	х	#DIV/0!	х	0.07	
	1234yf		0.06	#VALUE	<u>A2L</u>		#VALUE		0.07	
	236fa	C3H2F6, 1,1,1,3,3,3-Hexafluoropropane	2.72		A1				0.06	
	23	CHF3 Fluoroform	45.98		A1			35.3	0.03	
	125	C2HF5 Pentafluoroethane	13.90		A1			x	0.02	-
	227ea	C3HF7, 1,1,1,2,3,3,3-Heptafluoropropane	4.55		A1				0.01	
	*less	common ones in BACKUP	Δ1	R1. n	on_flan	nmabl	0			1

less common ones in BACKUP

A1, B1: non-flammable A2(L): low flammability

0.35 0.7 l 0.75 1.4 b	bar C2H bar C2H3 bar C1H bar C3H3	index HFC by A2(L)-towering 14F2 (LFL) 3F3 (LFL) 12F2 (remaining Q) 8F5 (v.p.)		flamma blility lower limit %	fraction of Q at LFL	flammab ility upper limit in	H-in dex= f/b * #H		
		2F4 (v.p.) 2F6 (remaining fraction)		4	8%	75	999.0		
		14F2[7]C2H3F3[8]C1H2F2[14]C3H3F5[6	BH2F6	<u>1.4</u>	36%	7.5	4.80		
	49.80					<u>1.4</u>	7376%	8.3	4.80
A H	101.21 2.28					<u>1.8</u>	40%	8.4	4.17
н В	2.28 47.52,					<u>1.8</u>	38%	8.5	4.17
	nent H 0	2.28, Element C 2 2.09, Element F	4 3.8	9		2.1	33%	9.5	3.56
purit	y: 0.0479	980				3	34%	12.5	3.00
H-in	dex: 0.1	09394				5	32%	15	2.67
	152a	C2H4F2 1,1-Difluoroethane	5.96	12.8%	A2	<u>3.7</u>	29%	18	0.53
	143a	C2H3F3 1,1,1-Trifluoroethane	12.62	16.0%	A2L	7	44%	<u>17</u>	0.23
	32	CH2F2 Difluoromethane	17.31	28.4%	A2L	13	46%	33	0.17
V I	245fa	C3H3F5, 1,1,1,3,3-Pentafluoropropane	1.48		B1				0.14
	134a	C2H2F4 1,1,1,2-Tetrafluoroethane, Norflurar	6.65		A1			х	0.08
1	1234ze	C3H2F4, 1,3,3,3-Tetrafluoropropene	7.00	#DIV/0!	<u>A2L</u>	х	#DIV/0!	х	0.07
1	1234yf	C3H2F4 2,3,3,3-Tetrafluoropropene	0.06	#VALUE	A2L		#VALUE	1	0.07
1	236fa	C3H2F6, 1,1,1,3,3,3-Hexafluoropropane	2.72		A1				0.06
	23	CHF3 Fluoroform	45.98		A1			35.3	0.03
	125	C2HF5 Pentafluoroethane	13.90		A1			x	0.02
1	227ea	C3HF7, 1,1,1,2,3,3,3-Heptafluoropropane	4.55		A1				0.01

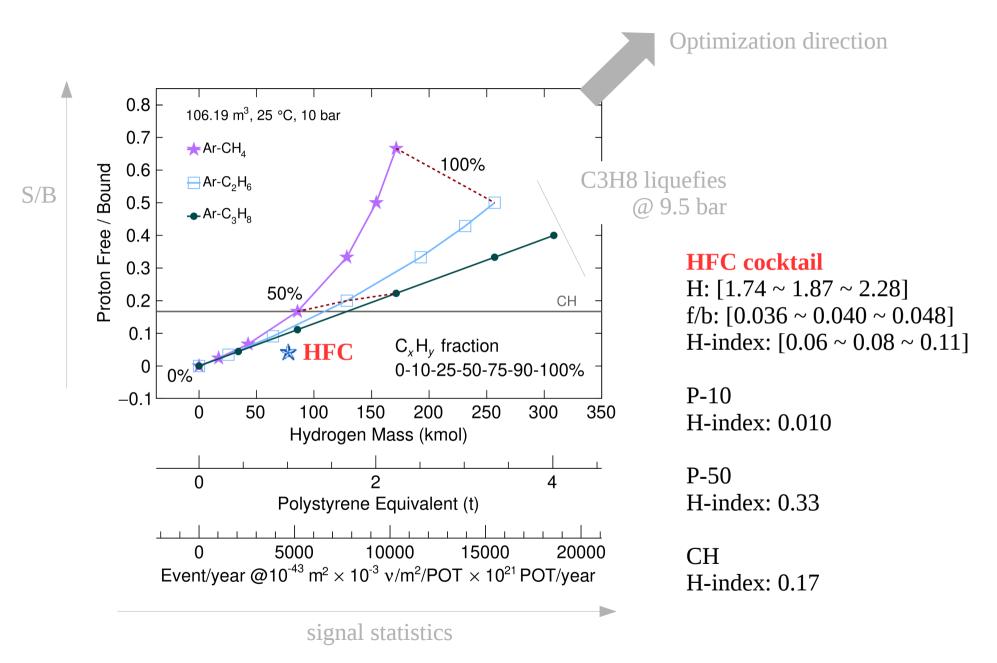
A1, B1: non-flammable A2(L): low flammability

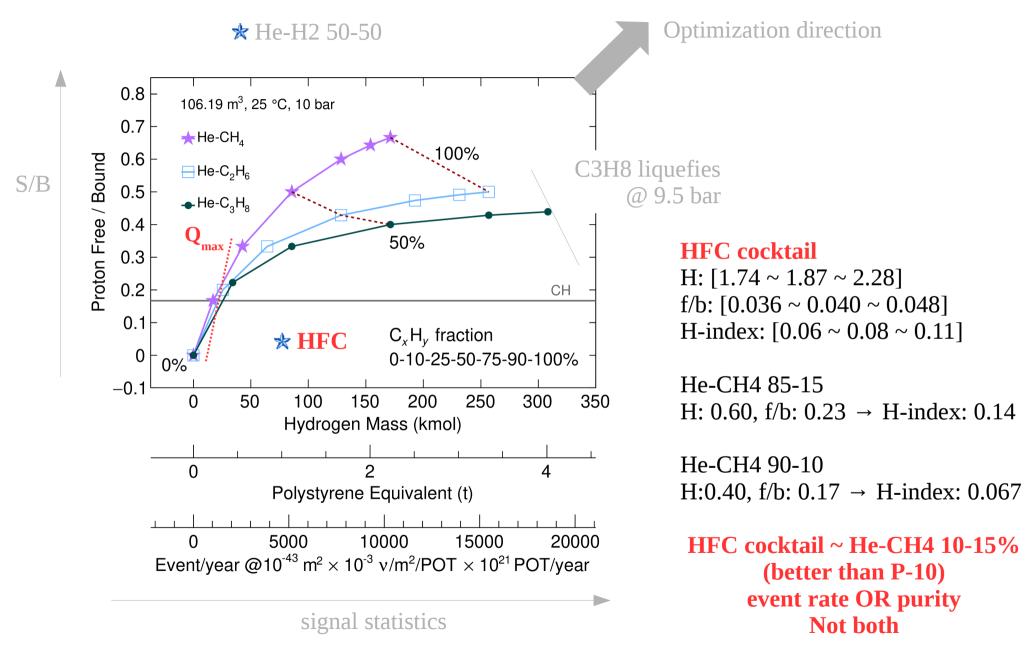
code name:allowedASHRA PercenthACv.p.percentE 34HFC-, R-25CageSafety group	flamma blility lower limit %	fraction of Q at LFL	flammab ility upper limit in	H-in dex= f/b * #H
All non-flammable HFC away from phase boundary	4	8%	75	999.0
0.7 bar C3H3F5	<u>1.4</u>	36%	7.5	4.80
4.5 bar C2H2F4 1.5 bar C3H2F6	<u>1.4</u>	7376%	8.3	4.80
3.3 bar CHF3	<u>1.8</u>	40%	8.4	4.17
Name [7]C3H3F5[45]C2H2F4[15]C3H2F6[33]C1H1F3	<u>1.8</u>	38%	8.5	4.17
Z 49.44	2.1	33%	9.5	3.56
A 101.18 H 1.74	3	34%	12.5	3.00
B = 47.70,	5	32%	15	2.67
Element H 0 1.74, Element C 2 1.89, Element F 4 4.04	<u>3.7</u>	29%	18	0.53
purity: 0.036478	7	44%	<u>17</u>	0.23
H-index: 0.063472	<u>13</u>	46%	33	0.17
245fa C3H3F5, 1,1,1,3,3-Pentafluoropropane 1.48 B1				0.14
134a C2H2F4 1,1,1,2-Tetrafluoroethane, Norflurar 6.65 A1			x	0.08
1234ze C3H2F4, 1,3,3,3-Tetrafluoropropene 7.00 #DIV/0! A2L	х	#DIV/0!	х	0.07
1234yf C3H2F4 2,3,3,3-Tetrafluoropropene 0.06 #VALUE A2L		#VALUE	:!	0.07
236fa C3H2F6, 1,1,1,3,3,3-Hexafluoropropane 2.72 A1				0.06
23 CHF3 Fluoroform 45.98 A1			35.3	0.03
125 C2HF5 Pentafluoroethane 13.90 A1			x	0.02
227ea C3HF7, 1,1,1,2,3,3,3-Heptafluoropropane 4.55 A1				0.01

A1, B1: non-flammable A2(L): low flammability

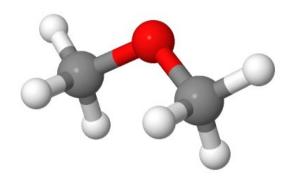
	code name: le High- oar CH2I	LFL Flammable Cocktail	v.p.	allowed percent	ASHRA E 34	flamma blility lower limit %	fraction of Q at LFL	flammab ility upper limit in	H-in dex= f/b * #H
	bar CH2F			4	8%	75	999.0		
4.5 t	oar C2H2	2F4		1.4	36%	7.5	4.80		
	oar C3H2			<u>1.4</u>	7376%	8.3	4.80		
2 bai Nam		3 LH2F2[7]C3H3F5[45]C2H2F4[15]C3H2F6	5[20]C1	H1F3		<u>1.8</u>	40%	8.4	4.17
Z	48.40		/[=0]01			<u>1.8</u>	38%	8.5	4.17
А	98.84					2.1	33%	9.5	3.56
H B	1.87					3	34%	12.5	3.00
	46.53, nent H 0	1.87, Element C 2 1.89, Element F	4 3.9)1		5	32%	15	2.67
	ty: 0.040			-		<u>3.7</u>	29%	18	0.53
H-in	idex: 0.0	75154				7	44%	<u>17</u>	0.23
	32	CH2F2 Difluoromethane	17.31	28.4%	A2L	<u>13</u>	46%	<u>33</u>	0.17
	245fa	C3H3F5, 1,1,1,3,3-Pentafluoropropane	1.48		B1				0.14
	134a	C2H2F4 1,1,1,2-Tetrafluoroethane, Norflurar	6.65		A1			х	0.08
	1234ze	C3H2F4, 1,3,3,3-Tetrafluoropropene	7.00	#DIV/0!	<u>A2L</u>	х	#DIV/0!	х	0.07
	1234yf		0.06	#VALUE	<u>A2L</u>		#VALUE	1	0.07
	236fa	C3H2F6, 1,1,1,3,3,3-Hexafluoropropane	2.72		A1				0.06
	23	CHF3 Fluoroform	45.98		A1			35.3	0.03
	125	C2HF5 Pentafluoroethane	13.90		A1			x	0.02
	227ea	C3HF7, 1,1,1,2,3,3,3-Heptafluoropropane	4.55		A1				0.01

A1, B1: non-flammable A2(L): low flammability

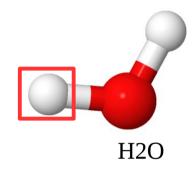




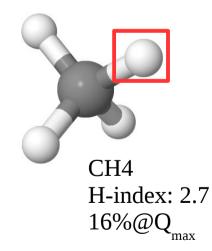
Search for H-rich non-flammable gas: CHO

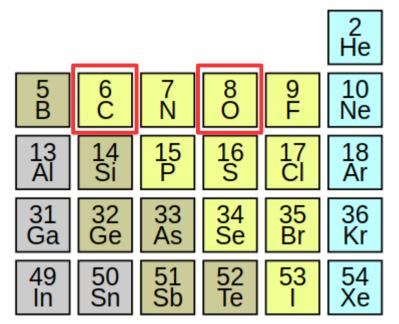


CH3OCH3 Dimethyl ether (DME) v.p. 5.9 bar H-index: 1.8 Flammable 3.4-18% (A3) 9.4%@Q_{max}



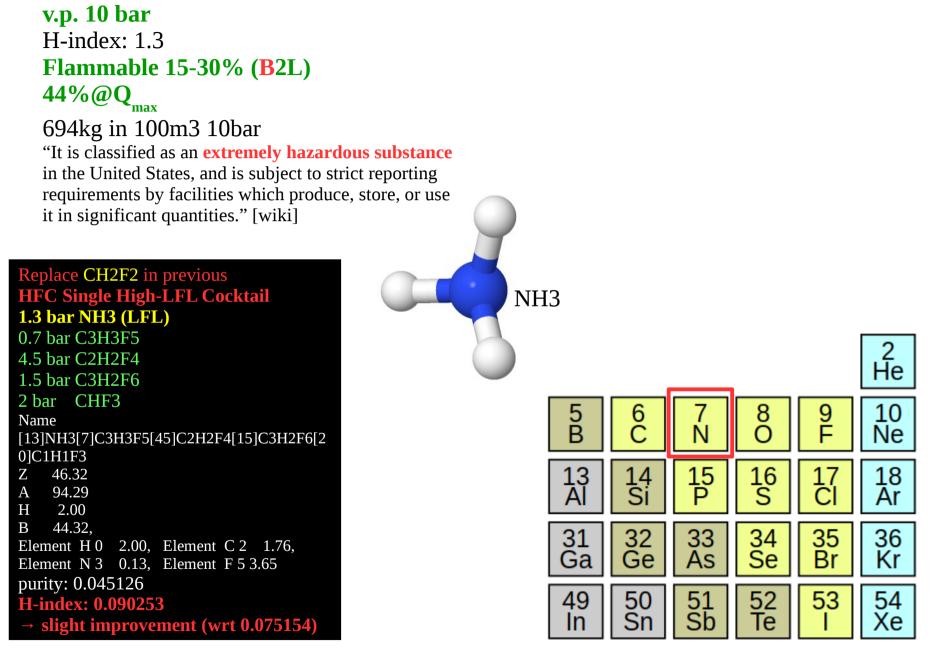
Any other CHO gas?





https://chemapps.stolaf.edu/jmol

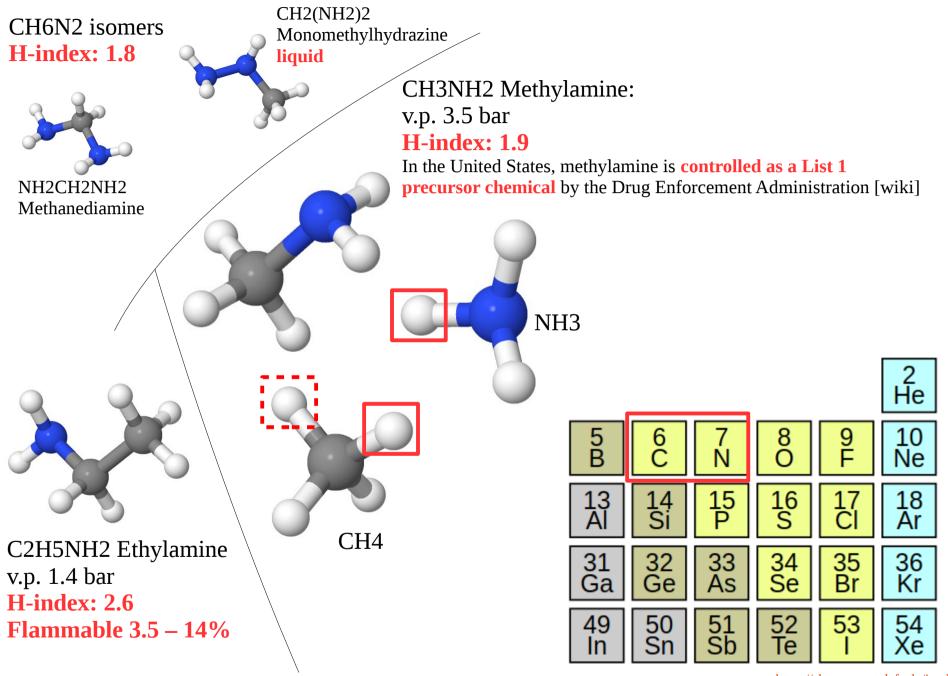
Search for H-rich non-flammable gas: CHN



Xianguo Lu, Oxford

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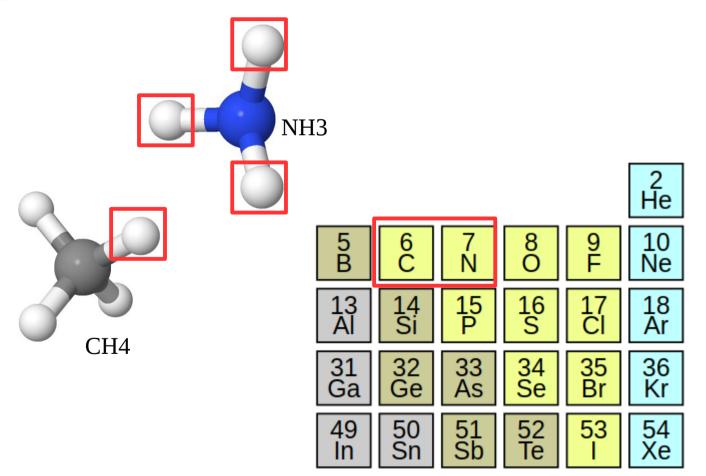
Xianguo Lu, Oxford

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Search for H-rich non-flammable gas: CHN

N(CH3)3 Trimethylamine (TMA) **v.p. 2.2 bar H-index: 3.2** (between C2H6 and C3H8) **Flammable 2 – 11.6% (too low LFL)** At lower concentrations, it has a "fishy" odor, the odor a

At lower concentrations, it has a "fishy" odor, the odor associated with rotting fish (wiki) Xe-TMA 6.2% at 10 bar (> LFL?!) JINST 8 (2013) P01012, arXiv:1210.3287



Summary and discussions (1)

code name: HFC-, R-		v.p. 25C (bar)	allowed percent age	ASHRA E 34 Safety group	flamma blility lower limit %	fraction of Q at LFL	flammab ility upper limit in	H-in dex= f/b * #H
	H2	superc	49.2%	A3	4	8%	75	999.0
	C5H12, Neope	1.72	3.8%	A3?	<u>1.4</u>	36%	7.5	4.80
	C5H12, isopar	0.92	0.0%	A3	<u>1.4</u>	7376%	8.3	4.80
600a	iC4H10, CH(C	3.51	4.5%	A3	<u>1.8</u>	40%	8.4	4.17
	C4H10, butane	2.43	4.7%	A3	<u>1.8</u>	38%	8.5	4.17
	C3H8	9.52	6.3%	A3	2.1	33%	9.5	3.56
	C2H6	41.90	8.9%	A3	3	34%	12.5	3.00
	CH4	superc	15.8%	A3	5	32%	15	2.67
	CH3OCH3, sir	5.90	9.4%	A3	<u>3.4</u>	36%	18	1.80
	NH3	10.03	43.9%	B2L	<u>15</u>	34%	30	1.29
152a	C2H4F2 1,1-D	5.96	12.8%	A2	<u>3.7</u>	29%	18	0.53
143a	C2H3F3 1,1,1	12.62	16.0%	A2L	7	44%	<u>17</u>	0.23
32	CH2F2 Difluor	17.31	28.4%	A2L	<u>13</u>	46%	<u>33</u>	0.17
245fa	C3H3F5, 1,1,1	1.48		B1				0.14
134a	C2H2F4 1,1,1	6.65		A1			x	80.0
1234ze	C3H2F4, 1,3,3	7.00	#DIV/0!	<u>A2L</u>	х	#DIV/0!	х	0.07
1234yf		0.06	#VALUE	A2L		#VALUE	!	0.07
236fa	C3H2F6, 1,1,1	2.72		A1				0.06
23	CHF3 Fluorofo	45.98		A1			35.3	0.03
125	C2HF5 Pentaf	13.90		A1			x	0.02
227ea	C3HF7, 1,1,1,1,2	4.55		A1				0.01

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Flammability hierarchy A3 \rightarrow (B \rightarrow A)2 \rightarrow (B \rightarrow A)1

Corresponding H-index hierarchy A2: below 1 A1: below 0.1

Summary and discussions (1)

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1234yf		0.06	#VALUE	<u>A2L</u>		#VALUE	!	0.07
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23	CHF3 Fluorofc	45.98		A1			35.3	0.03
125	C2HF5 Pentaf	13.90		A1			x	0.02
227ea	C3HF7, 1,1,1,1	4.55		A1				0.01

At lower flammability limit, most of the gas use up only 30-40% $Q_{_{\rm max}}$

Assume LFL constant (generally not true):

15% NH3 + 13% CH2F2 ~ 80% Q_{max} Leaving 7.2 bar 20% Q_{max} to fill up

Summary and discussions (2)

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UPDATE

Maximal H Cocktail 1.7 bar neopentane (C5) 3.5 bar isobutane (C4) 2.4 bar butane (C4) 2.4 bar propane (C3) Name [17]C5H12[59]C4H10[24]C3H8 Ζ 33.44 А 57.02 Η 9.86 В 23.58, Element H 0 9.86, Element C 2 3.93 purity: 0.418151 H-index: 4.122969

Summary and discussions (2)

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227ea	C3HF7, 1,1,1,1	4.55		A1				0.01

Single High-LFL Flammable HFC Cocktail 1.3 bar CH2F2 (LFL) 0.7 bar C3H3F5 4.5 bar C2H2F4 1.5 bar C3H2F6 2 bar CHF3 Name [13]C1H2F2[7]C3H3F5[45]C2H2F4[15]C3H2F6[20]C1H1F3 Z 48.40 A 98.84 H 1.87 B 46.53, Element H 0 1.87, Element C 2 1.89, Element F 4 3.91 purity: 0.040189 H-index: 0.075154

Summary and discussions (3)

- 1. Survey on HFC
 - Refrigerant industry
 - HFC has quite some non-flammable gas, but all with low H-index (necessarily)
 - One particular A2L (low flammability low toxicity) gas with high LFL (lower flammability limit):
 - > CH2F2 Difluoromethane, 13% LFL
 - Can be cooked for non-flammable mixtures
 - HFC cocktails ~ He-CH4 10-15%
 - Event rate OR purity, not both.
 - > Need to consider realistic tracking and DUNE ND run plan
 - Only looked at saturated HFC, non-saturated ones are more H-efficient
 - C2H2F2 H-index: 0.13, better than C2H2F4 (0.08), worse than CH2F2 (0.17), very flammable, allowed fraction is 2% by Q
- 2. Probed CHN
 - NH3 (B2L: low flammability *high* toxicity) has the highest H-index (1.29) among all low flammability gas, and even higher LFL (15%) than CH2F2
 - Generally, NH2 is more H-efficient than F when replacing H in alkane, but need a new strategy to explore as the industry is changed (refrigerant → drug); **might have surprises**
- 3. Briefly looked at CHO
 - CHO is mostly liquid except DME (CH3OCH3) which is A3 (high flammability)
 - OH is between NH2 and F in terms of H-efficiency; surprises?

Summary and discussions (3)

4. Need to sync with FNAL safety people

- Regulation details: how much parametric overlap between FESHM 6020.3 and ASHRAE 34?
- Refine gas search strategy: safety vs. tracking vs. possible run plan

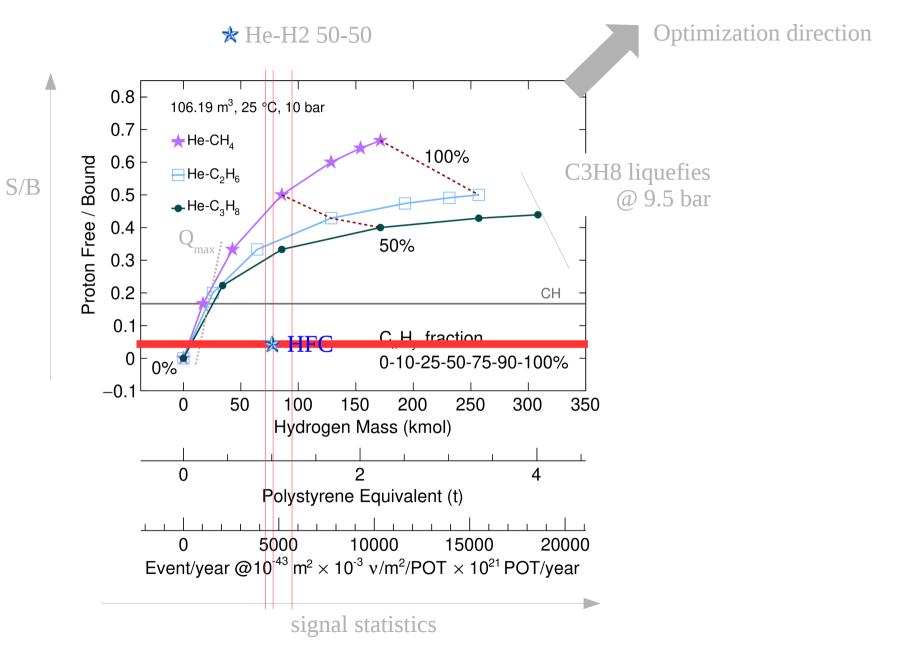
5. TPC gas properties benchmarking

- Ionization density can be large enough to enable μ/π separation outside dE/dx crossing?
- 3-body attachment at high pressure, etc.

BACKUP

code name: HFC-, R-		v.p. 25C (bar)	allowed percent age	ASHRA E 34 Safety group	flamma blility lower limit %	fraction of Q at LFL	flammab ility upper limit in	H-in dex= f/b * #H
41	CH3F Fluoromethane	36.64	#VALUE	x	<u>5.6</u>	#VALUE	22.2	0.60
161	C2H5F Fluoroethane	9.43	#VALUE	x	x	#VALUE	17.3	1.19
152	C2H4F2 1,2-Difluoroethane	7.18	#VALUE	x		#VALUE	!	0.53
		40.04		жх				0.13
143	C2H3F3 1,1,2-TRIFLUOROETHANE	x	14.5%	x		0%		0.23
134	C2H2F4 1,1,2,2-Tetrafluoroethane	x	30.4%	x		0%	x	0.08
281	C3H7F, 1-Fluoropropane	x	#DIV/0!	x		#DIV/0!		1.81
272	C3H6F2, Difluoropropane		#DIV/0!	x		#DIV/0!		1.00
263	C3H5F3, Trifluoropropane		#DIV/0!	x		#DIV/0!		0.56
254cb	C3H4F4, 1,1,2,2-Tetrafluoropropane		#DIV/0!	x		#DIV/0!		0.30
245eb	C3H3F5, 1,1,1,2,3-Pentafluoropropane		#DIV/0!	x		#DIV/0!		0.14

Search for H-rich non-flammable gas: HFC



END