

Fundamental wave-particle

relations: λ : wavelength

Frequency: $f = \frac{E}{h}$

$$\hbar = h/2\pi$$

wave number: $k = \frac{2\pi}{\lambda}$

angular frequency: $\omega = \frac{2\pi}{T}$

De Broglie

λ

$$p = \frac{h}{\lambda} = h \cdot 2\pi k = \hbar k$$

$$E = hf = \hbar \omega$$

FUNDAMENTAL WAVE PARTICLE
RELATIONSHIPS

For each wave, there is a wave equation of which the w.f. is a solution

$$\left\{ \begin{array}{l} \text{ex. electromagnetic wave, follow Maxwell equation} \\ \nabla \cdot \mathbf{E} = 0, \nabla \times \mathbf{E} = -\frac{\partial \mathbf{B}}{\partial t}; \nabla \cdot \mathbf{B} = 0; \nabla \times \mathbf{B} = \mu_0 \epsilon_0 \frac{\partial \mathbf{E}}{\partial t} \\ \text{classical wave eq. } \frac{\partial^2 u}{\partial t^2} = c^2 \frac{\partial^2 u}{\partial x^2} \end{array} \right.$$

Matter waves follow the Schrödinger equation:

- We consider a free particle (no external forces):

$$\text{Eq: } -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} \Psi(x,t) = i\hbar \frac{\partial}{\partial t} \Psi(x,t)$$

Meaning Born's statistical interpretation:

$|\Psi(x,t)|^2 dx$: probab. of finding a particle between $x, x+dx$ at time t

• FREE PARTICLE :

we guess a possible solution: $i(kx - \omega t)$

$$\text{PLANE WAVE } \Psi(x,t) = A e^{i(kx - \omega t)}$$

Is it a sol?

$$+\frac{\hbar^2}{2m} k^2 \left[A e^{i(kx - \omega t)} \right] = \hbar \omega A e^{i(kx - \omega t)}$$

$$\rightarrow \frac{\hbar^2 k^2}{2m} = \hbar \omega \rightarrow \frac{p^2}{2m} = E \quad \left\{ \begin{array}{l} \text{wave eq. f: } x \pm vt \\ i k(x - \omega t) \text{ this is a wave} \\ \text{ ; with } v = \frac{\omega}{k} = \sqrt{\frac{E}{2m}} \end{array} \right.$$

It is telling us that for a free particle, the schr. eq. is related to a classical accounting of ENERGY

The square of the wave function absolute value gives: probability/[unit of length] of finding the particle: $|\Psi(x,t)|^2$

SUMMARY

• FREE PARTICLE SCHRÖDINGER EQUATION:

$$-\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} \psi(x,t) = i\hbar \frac{\partial}{\partial t} \psi(x,t)$$

$$\psi(x,t) = A e^{i(kx - \omega t)} \quad \text{plane wave}$$

$$\frac{\hbar^2 k^2}{2m} = \hbar\omega$$

• FUNDAMENTAL PARTICLE-WAVE EQS

$$p = \frac{h}{\lambda} = \hbar k$$

$$E = h\nu = \hbar\omega$$

$$\psi(x,t) = A e^{\frac{i}{\hbar}(px - Et)}$$

PHYSICAL CONDITIONS: WELL BEHAVED WAVE FUNCTIONS

The probability of finding a particle between

$$x - x + dx: \quad P(x) dx = \psi^*(x,t) \psi(x,t) \cdot dx$$

PROBABILITY DENSITY

The probability of observing it between x_1 and x_2 is:

$$P = \int_{x_1}^{x_2} \psi^* \psi \cdot dx$$

- The particle must exist somewhere \rightarrow this condition is imposed NORMALIZING the w.f. [the extreme of integr. are $\neq \infty$]

$$\int_{-\infty}^{\infty} \psi(x,t)^* \psi(x,t) = 1$$

to represent a particle

e.g.

to be physically acceptable a w.f. must be normalizable

$$P.W. \int_{-\infty}^{\infty} dx |A|^2 \cdot e^{i(kx - \omega t)} \cdot e^{-i(kx - \omega t)} = \infty$$

* important property of schr. eq. is that it preserves the normaliz. at all times [Griffith eq. 1.121 and following]

\rightarrow we can understand it using the uncertainty principle. The problem with this wave is that cannot represent a particle whose w.f. is nonzero in a limited region of space. A localized wavefunction of this type is called wavepacket.

- SMOOTHNESS: w.f. smooth, 1 deriv. smooth, if not the k.E. could be ∞

What are expectation values:

For a particle in a state Ψ , the expectation value of x is:

$$\langle x \rangle = \int dx x \Psi^*(x) \Psi(x)$$

- what does it mean? It does not mean that if you measure the position of a particle over and over $\int x dx |\Psi|^2$ is the average of the results that you obtain

$\langle x \rangle$ is the average of measurements performed on particles All in state Ψ . Which means: - you prepare a whole ensemble of particles each in the same state Ψ and measure the positions of all of them $\langle x \rangle$ is then the average of these results

The expectation value is the average of repeated measurements on an ensemble of identically prepared systems; NOT the average of repeated measurements on the same system.

⊙ we say the operator \hat{x} "represents" position; to calculate its expectation value we sandwich the appropriate operator between Ψ and Ψ^* & integrate

$$\langle Q \rangle = \int dx \Psi^* Q \Psi$$

OPERATOR: is an instruction to do something on the function that follows it; the position operator tells you to multiply by 'x' the mom. oper. tells you to differentiate w respect to x .

Well Defined Observables: Eigenvalues

Given an operator \hat{Q} , the function $f(x)$ is an **eigenfunction** of the operator, if and only if the operator acting on the function yields a constant times the same function:

$$\hat{Q} \underbrace{f(x)}_{\text{eigenfunction}} = \underbrace{\lambda}_{\text{Eigenvalue}} f(x)$$

In general, an operator will have multiple eigenvalues

EIGENVALUES, EIGENVECTORS

* How does it work for matrices?

We say v is an eigenvector of M if: $Mv = \lambda v$
is a number, eigenvalue

Example: **POLL**

$$\begin{bmatrix} a_1 & 0 & 0 \\ 0 & a_2 & 0 \\ 0 & 0 & a_3 \end{bmatrix} \rightarrow \begin{bmatrix} 1 \\ 0 \\ 0 \end{bmatrix} \begin{bmatrix} 0 \\ 1 \\ 0 \end{bmatrix}, \begin{bmatrix} 0 \\ 0 \\ 1 \end{bmatrix}$$

what are the eigen values: $1, 1, 1$ (A)
 $0, 0, 0$ (B)
 a_1, a_2, a_3 (C)

Heisenberg Uncertainty Relation:

The mere fact that a phenomenon has a wave nature implies inherent uncertainty in its particle properties.

What do we mean by "uncertainty" in an observable?

That if we measure many times the same obs., the number I find varies in a range of values

UNCERTAINTY \rightarrow MEASURE HOW FAR DEVIATIONS ARE FROM THE MEAN AVERAGE VALUE

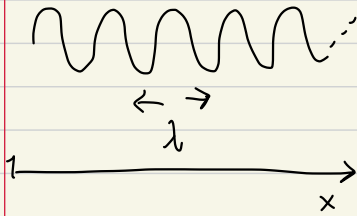
HEISENBERG UNCERTAINTY PRINCIPLE:

\rightarrow Because of a particle wave's nature, it is theoretically impossible to know precisely both its position along an axis and its momentum component along that axis

• Δx and Δp_x cannot be zero simultaneously

$$\Delta p_x \cdot \Delta x \geq \frac{\hbar}{2} \quad (\text{Heisenberg Nobel prize 1932})$$

Examples to better understand the uncertainty principles:

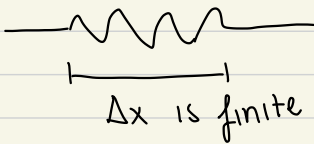


this wave is infinite and regular
well defined wavelength λ

$p = \frac{h\nu}{\lambda}$; Any idea on Δp ?

Δp_x is 0; but we don't know anything about its position

$\Delta x = \infty$

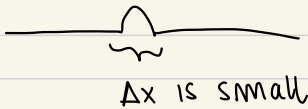


Wave is regular over a finite region
Not everywhere { λ and p not well defined }

$\Delta p \neq 0$

Better known position.

we further restricted the region where
the wave is regular and we can
define λ, p



HAMILTONIAN:

In classical mechanics the total energy:
kin + potential is called: Hamiltonian

$$\text{Hamiltonian} = \text{kin. en} + \text{pot. en}$$

$$H(x,p) = \frac{p^2}{2m} + V(x)$$

the Hamiltonian operator obtained by the canonical substit.
 $p \rightarrow -i\hbar \frac{\partial}{\partial x}$ is

$$\hat{H} = -\hbar^2 \frac{\partial^2}{\partial x^2} + V(x,t)$$

→ We saw the Schr. eq. for a free particle:

- $-\hbar^2 \frac{\partial^2}{\partial x^2} \psi(x,t) = i\hbar \frac{\partial}{\partial t} \psi(x,t)$

- general expression: $\hat{H} \psi(x,t) = i\hbar \frac{\partial}{\partial t} \psi(x,t)$

$$-\hbar^2 \frac{\partial^2}{\partial x^2} \psi(x,t) + V(x,t) \psi(x,t) = i\hbar \frac{\partial}{\partial t} \psi(x,t)$$

this is called TIME DEP. SCHRÖDINGER EQ

class of problems

→ STATIONARY STATES:

If the potential does not dep. on t ; then the Schr. eq can be solved by:

• $\Psi(x,t) = \psi(x) \cdot \phi(t)$ separation of variables ⎧ ① temporal
⎩ ② spatial

$$i\hbar \psi(x) \frac{\partial \phi(t)}{\partial t} = -\hbar^2 \phi(t) \frac{\partial^2 \psi(x)}{\partial x^2} + V(x) \psi(x) \phi(t)$$

we can div / $\psi(x)\phi(t)$

$$i\hbar \frac{1}{\phi(t)} \frac{\partial \phi(t)}{\partial t} = -\hbar^2 \frac{1}{\psi(x)} \frac{\partial^2 \psi(x)}{\partial x^2} + V(x)$$

each of these terms needs to be = constant

2 independent equations:

TEMPORAL PART: $i\hbar \frac{1}{\phi(t)} \frac{d\phi(t)}{dt} = C$ we denote it by E

• first order diff. equation: $\phi(t) = e^{-i\frac{E}{\hbar}t}$

$$\left(\frac{E}{\hbar}\right) = [t]^{-1} \underset{\sim \text{freq}}{=} \omega \rightarrow C = \hbar\omega = E$$

for the sp. part we solve: ^{this is called} TIME INDEP. SCHRÖD. eq

$$-\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} \Psi(x) + U(x) \Psi(x) = E \Psi(x)$$

$$\hat{H} \Psi(x) = E \Psi(x)$$

→ we want to find eigenvalues and eigenstates for our operator:

$$\hat{H}$$

the w.f. dep. on t

stationary state: $\Psi(x,t) = \Psi(x) e^{-\frac{iEt}{\hbar}}$

the prob. density, which is

Physical property: $|\Psi(x,t)|^2 = \Psi^*(x) \Psi(x)$ → does not dep. on t

for stationary states, the whereabouts of the particle do not change with time

Why stationary states are cool:

Classically an electron (charged particle) orbiting in an atom should constantly lose en. in the form of e.m. radiation.

In Q.M. the electron is a stationary cloud. its prob. density is = const., its charge density is = const. its energy is a well defined const. Q.M. explains atom stability
accel. charge radiates en. Larmor formula